

# NMR Training Course

9<sup>th</sup> September 2021  
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JEOL UK Demo Lab



# Power handling liquids probes

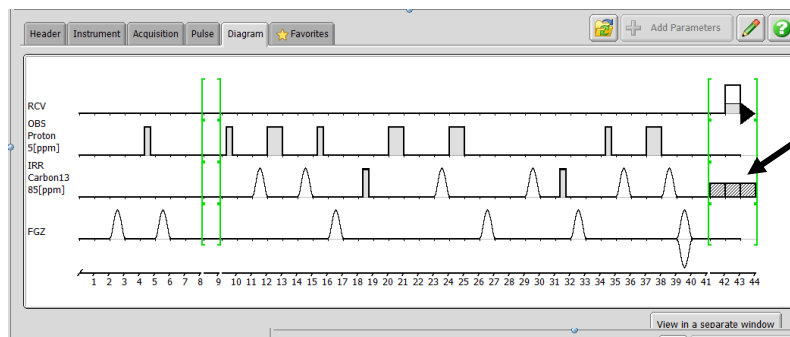
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# Power handling

- Probes can get damaged if they receive more RF than they can handle, this damage may extend to other elements in the spectrometer
- Clearly exceeding safe RF power levels will trigger an alarm in the RF amplifiers, but RF amplifiers can typically handle more power than the probes, so RF power levels below this limit may still damage the probe
- There is an absolute limit of maximum power that probes can tolerate at each frequency. RF irradiation beyond this level will likely damage the probe
- At lower power levels it is possible to irradiate longer. See following tables to evaluate how long can RF irradiation be at different power levels
- **Do not exceed the maximum irradiation time at each attenuation level**
- **In Delta a larger attenuation number [dB] results in lower RF power**
- **Maximum irradiation at Square pulse power is 1[ms] for HF and 0.2[ms] for LF**
- **Maximum decoupling power is 1.5W for 1H and 19F, and 5W for other nuclei. It is not guaranteed probe will survive if decoupling at this power level continuously.**
- The **maximum duty cycle** for decoupling at maximum power and with CPMG experiment types is: **5%**
- % is used for reference in solids.
- Power is used for reference, it varies with probes and nuclei. Typically maximum power for 1H is 30W and maximum continuous power is 1.5W. If different, the table should be modified accordingly
- Pulsewidth is used for reference on how it theoretically scales with attenuation

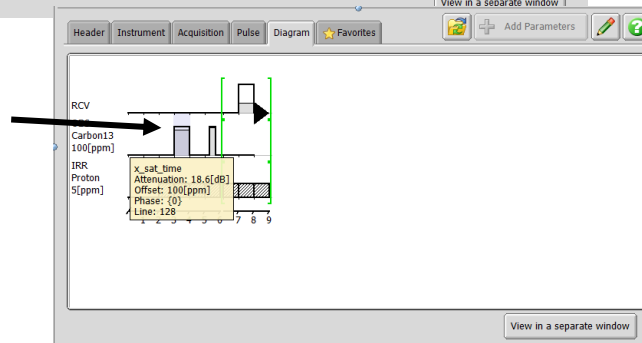
# Power handling

- **Power limits are safe limits, not an objective. Always use the lowest power level that allows you to obtain good results (if you want your probe to last longer).**
- **Always check power levels when using a pulse sequence for the first time**
- **Exceeding the power levels in these tables will void the warranty**
- **In case of doubt, ask!**
- **Check Diagram to quickly find any long RF irradiation**



Carbon 13 decoupling: Check irr\_atn\_dec, if less attenuation (more power) than CW, evaluate duty cycle: is acquisition time less than maximum irradiation time at this attenuation?

Long pulse (double line at the top): click on it to find out pulse length parameter, x\_sat\_time, and check its power level and duty cycle



# Power handling HF channel

Attenuation	power	irradiation	% (Voltage)	pulsewidth
x_atn+0[dB]	30W	1ms	100%	10us
x_atn+3[dB]	15W	5ms	70%	
x_atn+6[dB]	7.5W	25ms	50%	20us
x_atn+9[dB]	3.75W	125ms	35%	
x_atn+12[dB]	1.88W	625ms	25%	40us
x_atn+15[dB]	1W	CW	17.7%	

These are the maximum irradiation times with a duty cycle of 2%, at the specified attenuations (x\_atn= attenuation level for the 90 pulse in probe specifications)

In general, do not use attenuation levels below **x\_atn+18[dB]** for **decoupling**

[This table assumes attenuation changes linearly and irradiation is then calculated from Square pulse attenuation; but amplifiers are not so linear at the highest power, so a safety margin is recommended]

# Power handling LF channel

Attenuation	power	irradiation	% (Voltage)	pulsewidth
x_atn+0[dB]	160W	0.2ms	100%	10us
x_atn+3[dB]	80W	1 ms	70%	
x_atn+6[dB]	40W	5 ms	50%	20us
x_atn+9[dB]	20W	25ms	35%	
x_atn+12[dB]	10W	125ms	25%	40us
x_atn+15[dB]	5W	625ms	17.7%	
x_atn+18[dB]	2W	CW	12.5%	80us

These are the maximum irradiation times with a duty cycle of 2%, at the specified attenuations (x\_atn= attenuation level for the 90 pulse in probe specifications)

In general, do not use attenuation levels below **x\_atn+21[dB]** for **decoupling**

[This table assumes attenuation changes linearly and irradiation is then calculated from Square pulse attenuation; but amplifiers are not so linear at the highest power, so a safety margin is recommended]

# 400 ROYAL specifications

LCT	Maximum power		Maximum continuous power	
	W	dB	W	dB
	1H	30		
	19F	30	1.5	
X	31P	50	5	
X	11B	50	5	
A	13C	80	5	
A	29Si	80	5	
A	2H	80	5	
B	17O	150	5	
C	15N	150	5	
D	39K	80	5	

## Other nuclei

Lithium7, Tin117, Tin119

Bromine81, Copper63, Copper65, Gallium71, Praseodymium141, Rubidium87, Sodium23, Tellurium125, Vanadium51, Xenon129

Aluminum27, Bromine79, Europium151, Gallium69, Manganese55, Niobium93, Scandium45, Tellurium123, Terbium159

Antimony121, Cadmium111, Cadmium113, Cobalt59, Holmium165, Indium113, Indium115, Iodine127, Lead207, Platinum195, Rhenium185, Rhenium187, Technetium99

Arsenic75, Bismuth209, Lithium6, Mercury199, Selenium77, Ytterbium171

Antimony123, Beryllium9, Cesium133, Lanthanum138, Lanthanum139, Lutetium175, Tantalum181

Boron10, Barium137, Europium153

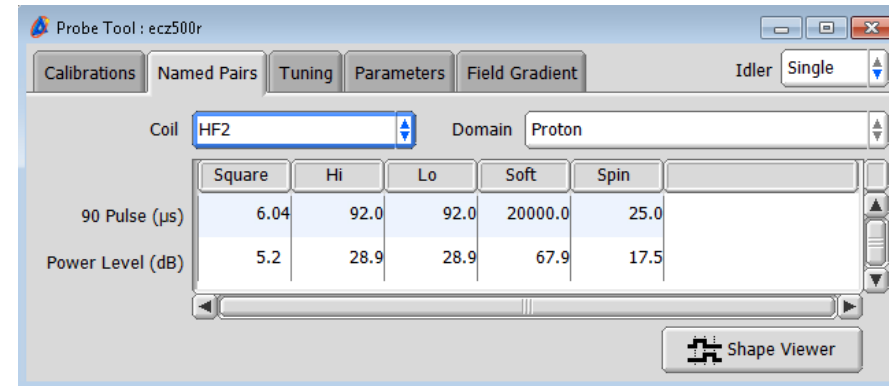
Silver109

Lock	15		-
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Measured

# Probe tool

- Named pairs tab contains the calibrations for each nucleus
- Square is used for hard pulses
- Hi is used for decoupling during acquisition
- Lo is used for decoupling during relaxation delay
  - Normally Hi/Lo pulses are set lower than CW decoupling for safety
  - \*Note Wurst(xx) and BUSS decouplings do not use Hi/Lo decoupling
- Soft is used for shaped pulses
  - \*\*There may be some exceptions
- Spin is used for spin-lock pulses (TOCSY, saturation-recovery, etc.)
- Nuclei not calibrated appear with 1[us] 90 pulse at 79[dB]
- Note different calibrations for Proton and Fluorine in the HFX probe. Idler can be Single or Dual





# Power handling solids

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# Power handling AutoMAS 3.2 400 MHz

## 1H (HF) channel

Attenuation	power	irradiation	% (Voltage)	pulsewidth
x_atn+0[db]	80W	50 ms	100%	2.7 us
x_atn+3[db]	40W	100 ms	70%	
x_atn+6[db]	20W	200 ms	50%	5.4 us

## 13C (LF) channel

Attenuation	power	irradiation	% (Voltage)	pulsewidth
x_atn+0[db]	220W	5 ms	100%	2.5 us
x_atn+3[db]	110W	10 ms	70%	
x_atn+6[db]	55W	20 ms	50%	5.0 us

## Maximum 1.25 % duty cycle

	<sup>1</sup> H/ <sup>19</sup> F	<sup>31</sup> P/ <sup>7</sup> Li/ <sup>11</sup> B	<sup>23</sup> Na/ <sup>27</sup> Al/ <sup>13</sup> C	<sup>207</sup> Pb	<sup>29</sup> Si	<sup>6</sup> Li	<sup>15</sup> N
Maximum decoupling power(W):	80	176	220	248	248	256	272

Maximum input power: Maximum decoupling power \* 1.25

# Power handling HXMAS 4mm 400 MHz

## 1H (HF) channel

Attenuation	power	irradiation	% (Voltage)	pulsewidth
x_atn+0[db]	48W	80 ms	100%	3.9 us
x_atn+3[db]	24W	160 ms	70%	
x_atn+6[db]	12W	320 ms	50%	7.8 us

## 13C (LF) channel

Attenuation	power	irradiation	% (Voltage)	pulsewidth
x_atn+0[db]	144W	5 ms	100%	3.7 us
x_atn+3[db]	72W	10 ms	70%	
x_atn+6[db]	36W	20 ms	50%	7.4 us

## Maximum 1.6 % duty cycle

	<sup>1</sup> H/ <sup>19</sup> F	<sup>31</sup> P/ <sup>7</sup> Li/ <sup>11</sup> B	<sup>23</sup> Na/ <sup>27</sup> Al/ <sup>13</sup> C	<sup>207</sup> Pb	<sup>29</sup> Si	<sup>6</sup> Li	<sup>15</sup> N
Maximum decoupling power(W):	48	112	144	160	160	184	224

Maximum input power: Maximum decoupling power \* 1.25

# AutoMAS 3.2mm vs HXMAS 4mm

- AutoMAS

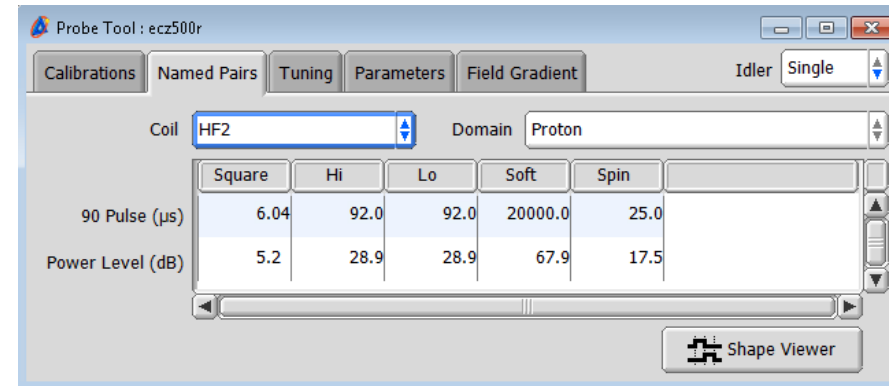
Resolution:	0.1 ppm or lower (specified at ADM $^{13}\text{C}$ FWHM)
Sensitivity:	120 or higher (specified at HMB $^{13}\text{C}$ 8 scans)
Maximum MAS speed (at room temperature):	22,000 Hz (Guaranteed speed: 21,000 Hz)
Gradient field:	Not included
Variable temperature range:	-60°C to +150°C

- HXMAS

Resolution:	0.05 ppm or lower (ADM, FWHM of $^{13}\text{C}$ )
Sensitivity:	260 or higher (HMB, 8 scans, S/N of $^{13}\text{C}$ )
Maximum MAS speed at room temperature	18,000 Hz (guaranteed 17,000 Hz)
Field gradient:	Not available
Variable temperature range:	-20 °C to +80 °C

# Probe tool

- Named pairs tab contains the calibrations for each nucleus
- Square is used for hard pulses
- ~~• Hi is used for decoupling during acquisition~~
- ~~• Lo is used for decoupling during relaxation delay~~
  - ~~— Normally Hi/Lo pulses are set lower than CW decoupling for safety~~
  - ~~— \*Note Wurst(xx) and BUSS decouplings do not use Hi/Lo decoupling~~
- ~~• Soft is used for shaped pulses~~
  - ~~— \*\*There may be some exceptions~~
- ~~• Spin is used for spin lock pulses (TOCSY, saturation-recovery, etc.)~~
- Nuclei not calibrated appear with 1[us] 90 pulse at 79[dB]
- ~~• Note different calibrations for Proton and Fluorine in the HFX probe. Idler can be Single or Dual~~



In standard solids experiments only the Square calibration is used.

CP, decoupling or other calibrations are not saved in probe tool

## Standard Tube

NM-05420ST4 (4mm) / NM-05410ST32 (3.2mm)

Items	Materials	
Spinning Cap	VespeI ®*1	
Bottom Cap	PEEK *2	
Sleeve (Color : White)	Zirconia (ZrO <sub>2</sub> )	
Spacer	PCTFE*3	

\*1: VespeI ®: A trademark or registered trademark of E.I. du Pont de Nemours and Company or its affiliates.

\*2:PEEK: PolyEther Ether Ketone

\*3:PCTFE: Poly Chloro Tri Fluoro Ethylene

### Spinning Detection Maker : Black permanent ink pen

		NM-05420ST4 (4mm)	NM-05410ST32 (3.2mm)
Maximum MAS speed (Guaranteed MAS speed)	At room temperature	18,000 Hz (17,000 Hz)	22,000 Hz (21,000Hz)
	At high temperature VT	16,000 Hz (15,000 Hz)	19,000 Hz (18,000 Hz)
	At low temperature VT	14,000 Hz (13,000 Hz)	17,000 Hz (16,000 Hz)
Variable Temperature Range	with spacer	-100 to +80 °C	
	without spacer	-100 to +150 °C	

The MAS speed of the sample tube is limited by changing the temperature.

The usable temperature range varies depending on the presence or absence of spacers.

## Heat Resisting Tube

NM-05470HST4 (4mm) / NM-05460HST32 (3.2mm)

Items	Materials	
Spinning Cap	Vespel ®*1	
Bottom Cap	PEEK *2	
Sleeve (Color : Yellow)	Heat Resisting Zirconia (ZrO <sub>2</sub> )	
Spacer	PTFE*3	


\*1:Vespel ®: A trademark or registered trademark of E.I. du Pont de Nemours and Company or its affiliates.

\*2:PEEK: PolyEther Ether Ketone

\*3:PTFE: Poly Tetra Fluoro Ethylene

### Spinning Detection Maker : Black permanent ink pen

		NM-05470HST4 (4mm)	NM-05460HST32 (3.2mm)
Maximum MAS speed (Guaranteed MAS speed)	At room temperature	18,000 Hz (17,000 Hz)	22,000 Hz (21,000Hz)
	At high temperature VT	16,000 Hz (15,000 Hz)	19,000 Hz (18,000 Hz)
	At low temperature VT	14,000 Hz (13,000 Hz)	17,000 Hz (16,000 Hz)
Variable Temperature Range		-100 to +200 °C	

 The MAS speed of the sample tube is limited by changing the temperature.

## Sealing Tube


NM-05450SST4 (4mm) / NM-05440SST32 (3.2mm)

Items	Materials	
Spinning Cap	PCTFE*1	
Bottom Cap	PCTFE*1	
Sleeve (Color : White)	Zirconia (ZrO <sub>2</sub> )	
Spacer	PCTFE*1	

\*1 : PTFE: Poly Tetra Fluoro Ethylene

### Spinning Detection Maker : Gold permanent ink pen

		NM-05450SST4 (4mm)	NM-05440SST32 (3.2mm)
Maximum MAS speed (Guaranteed MAS speed)	At room temperature	9,000 Hz (8,000 Hz)	10,000 Hz (9,000Hz)
	At high temperature VT	6,500 Hz	8,000 Hz
Variable Temperature Range		-20 to +50 °C	

 The MAS speed of the sample tube is limited by changing the temperature.

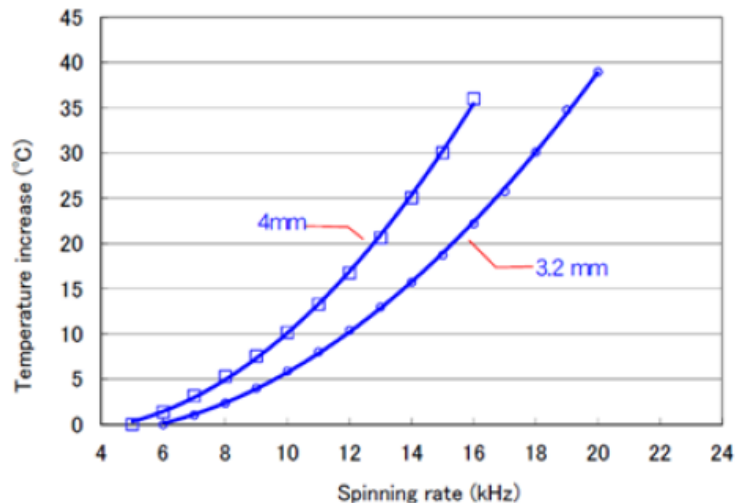


## ■ Too much a sample

If too much sample in the solid sample tube, the cap cannot close completely.

If the sample tube is spinning while the cap is not completely closed, the cap may remove out during spinning, the probe causes to damage.

Especially, solid-state NMR increases the temperature of the sample itself by spinning the sample tube at high speed.



KBr can be used as a thermometer.  
 $\Delta T = 0.025 \text{ ppm/K}$   
<https://dx.doi.org/10.1016/j.jmr.2008.09.019>

Generally, the sample volume increases due to thermal expansion as the sample temperature rises.

Although the coefficient of thermal expansion varies depending on the sample, if there are too many samples, the expanded sample pushes up the cap, which may cause the cap to come off the sleeve.

Therefore, it is better to fill a little less sample in the solid sample tube.

Thermal expansion

Glass  $\ll$  Metal  $\ll$  Resin

Metal is 4 times larger than glass, and resin is 20-30 times larger than glass.

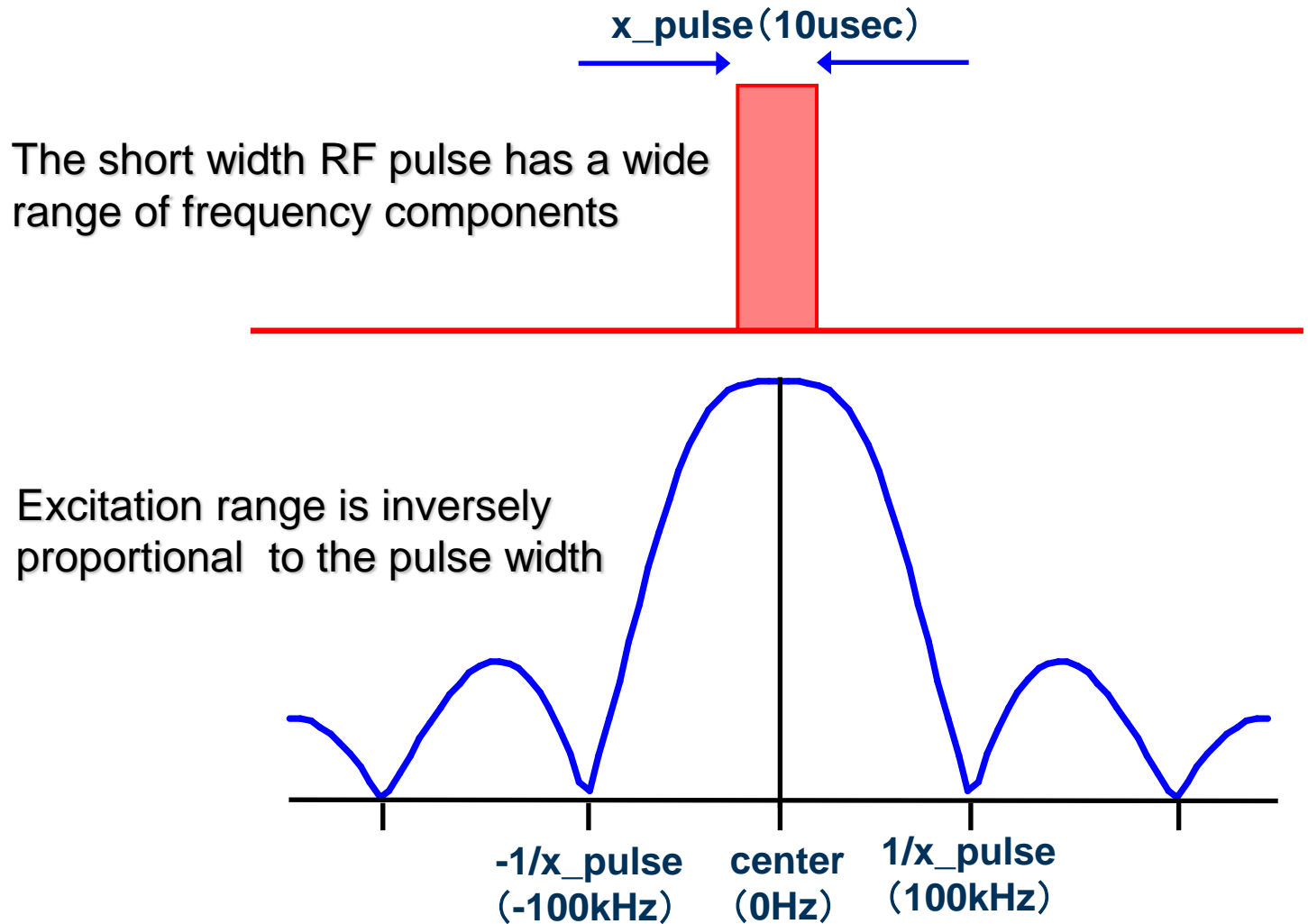
# Webinars

- An Introduction to Solid-State NMR  
<https://attendee.gotowebinar.com/register/1588889267810221067>
- Solid-State NMR Tutorial: Sample Packing, Standard Samples & Sample Spinning  
<https://attendee.gotowebinar.com/register/8621407423140093454>
- More webinars:  
[https://www.jeol.co.jp/en/news/seminar/webseminar/movie\\_index.html](https://www.jeol.co.jp/en/news/seminar/webseminar/movie_index.html)

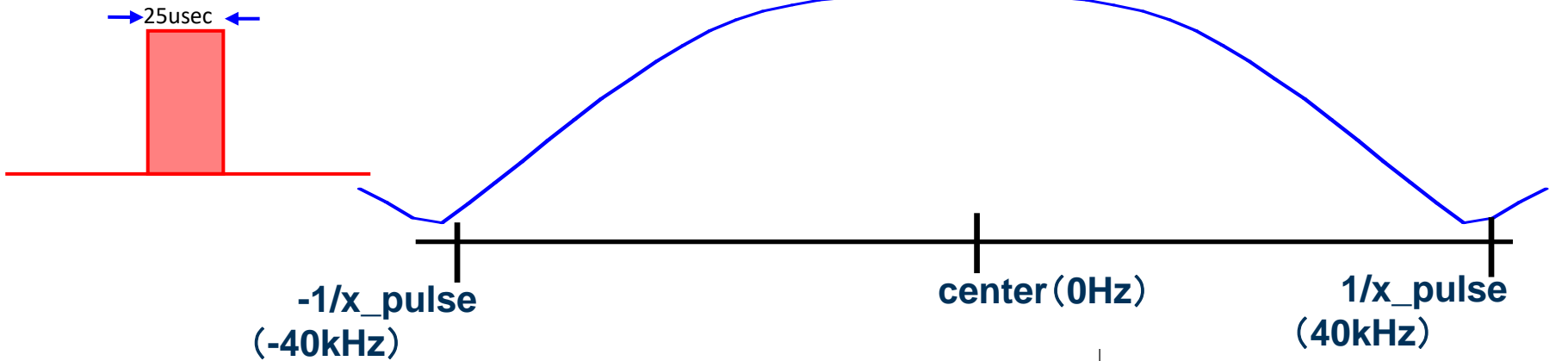
# Pulse hard vs soft

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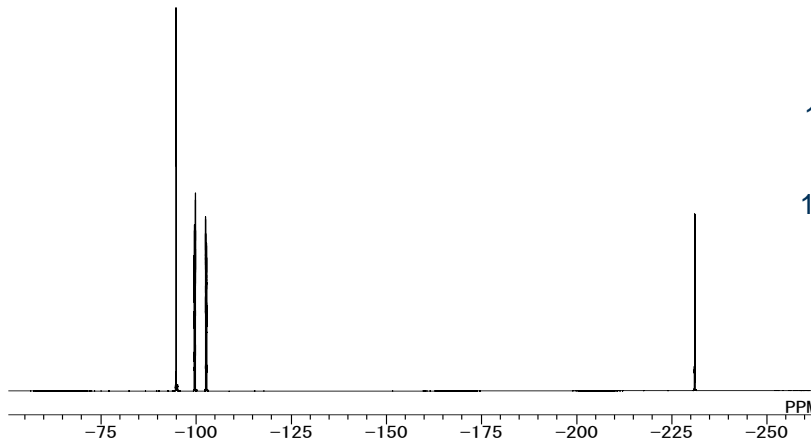
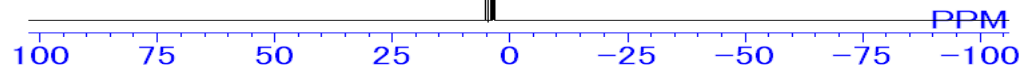
# Excitation profile of a 90 degree pulse



# Excitation profile of a 90 degree pulse



$^1\text{H}$  10ppm at a 400MHz machine is only 4kHz.  
A 25usec pulse can excite all  $^1\text{H}$  peaks equally.

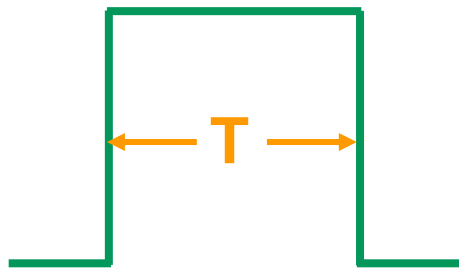


$^{19}\text{F}$  peak at -230ppm will be excited very weakly

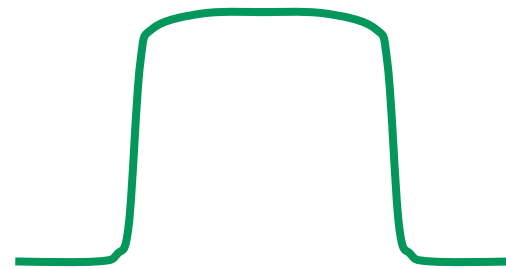
$^{19}\text{F}$  on a 400MHz instrument

# Rectangular pulses

- Referred to as “hard” pulses
- Effective at exciting broad bandwidths, with relatively flat excitation over the frequency range 0 to  $\sim 1/T$  Hz
- Shorter pulses lead to broader excitation ( $1/T$  is larger)
- Poor choice for selective (narrow frequency range) excitation (need very long  $t$ !)



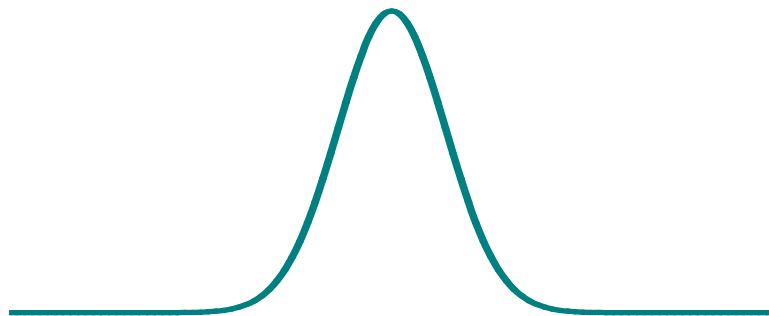
**Ideal rectangular**



**Actual rectangular**

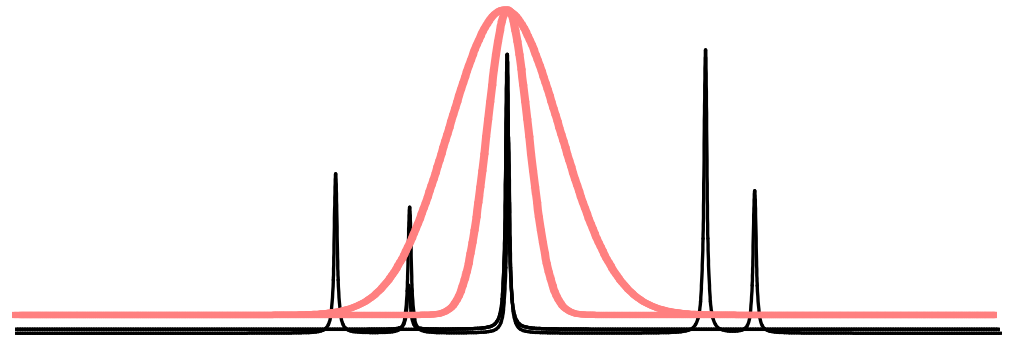
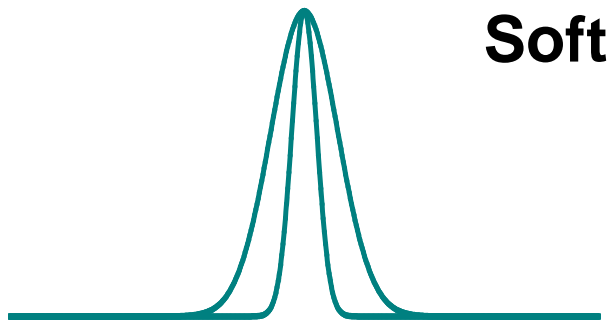
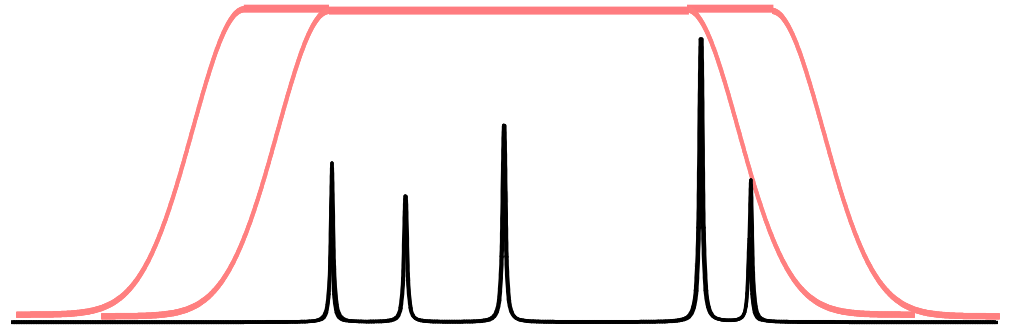
# Gaussian pulses

- Referred to as “soft” pulses
- Excitation profile is also Gaussian (narrow with relatively rapid falloff)
- Shorter pulses correspond to narrow excitation profiles
- Useful for selective pulses



**Gaussian**

# Hard vs. soft pulses

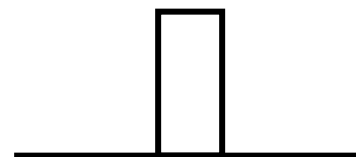




# Adiabatic pulses

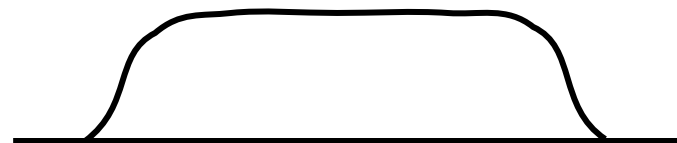
## Rectangular pulses

- + Very short pulse width ( $\mu\text{s}$ )
- Small excitation profile
- High power

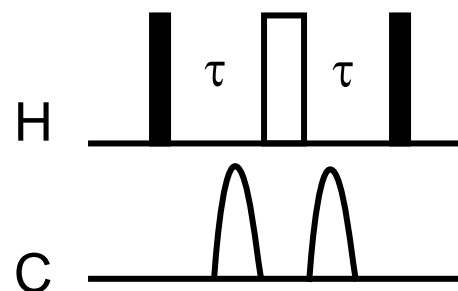


## Adiabatic pulses

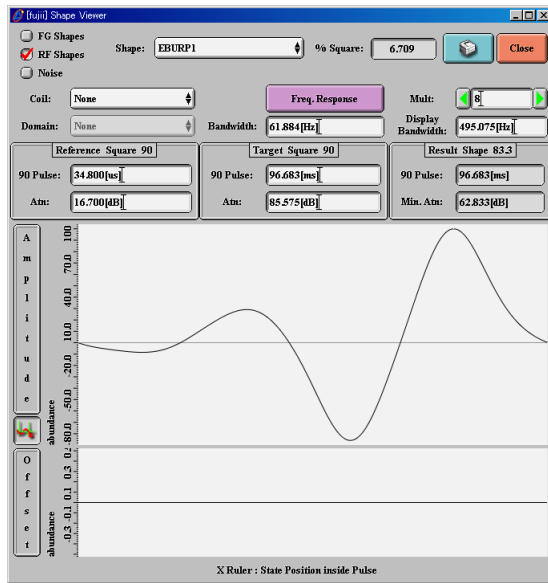
- Adiabatic pulses do frequency sweep of kHz to MHz, spin magnetization follows the slow (adiabatic) inversion profile
- Replace bandwidth-critical  $180^\circ$  pulses by adiabatic pulses
- + Extremely wide excitation bandwidth (essential for fields  $\geq 600$  MHz!)
- + Low power
- Long pulses (typ. 1-50 ms) have to be compensated in pulse sequences



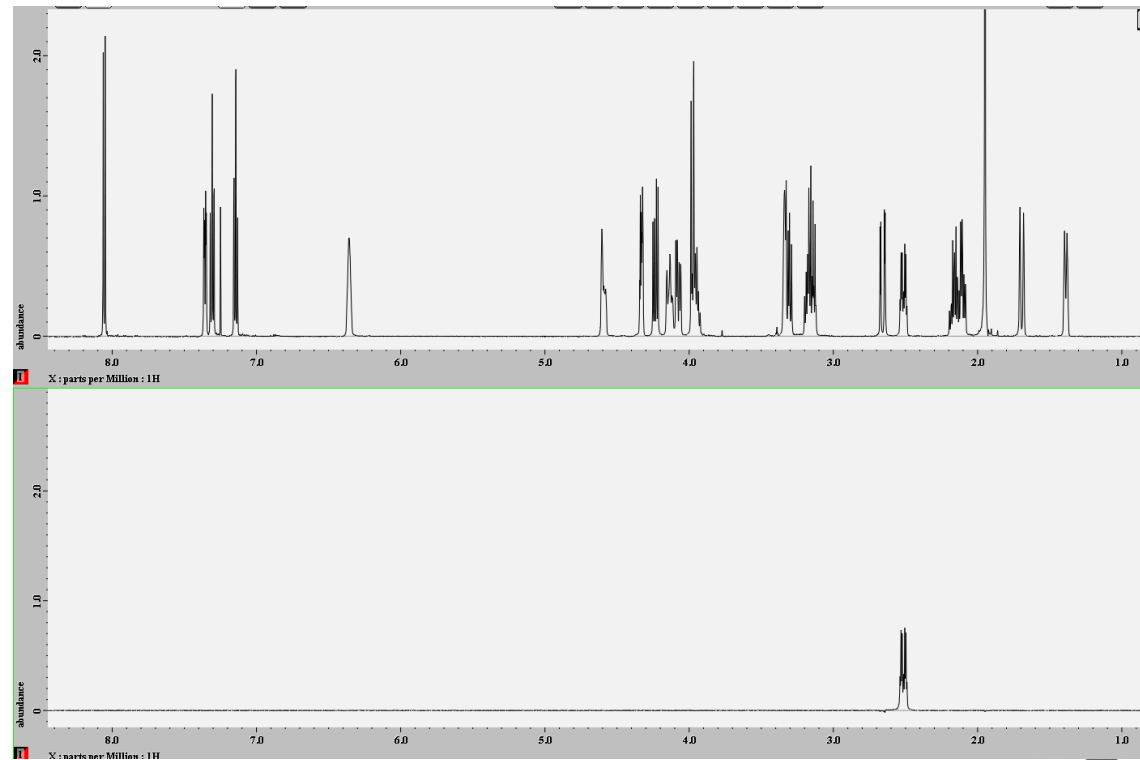
## adiabatic INEPT



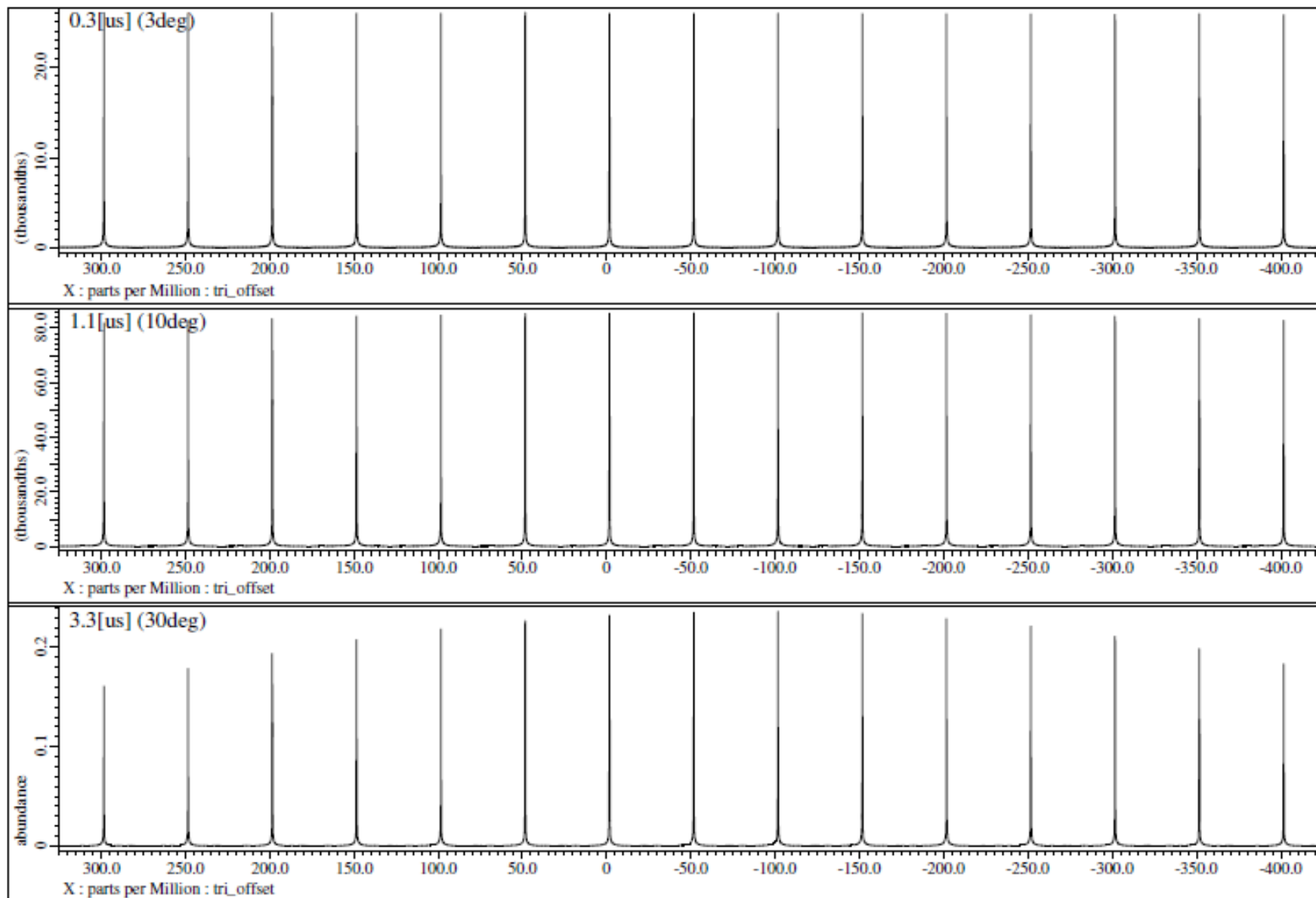
# Example of shaped pulse



96msec EBURP pulse time domain shape (upper) and a result (right lower)



# Fluorine excitation (hard pulse in a ECZ-S 400)

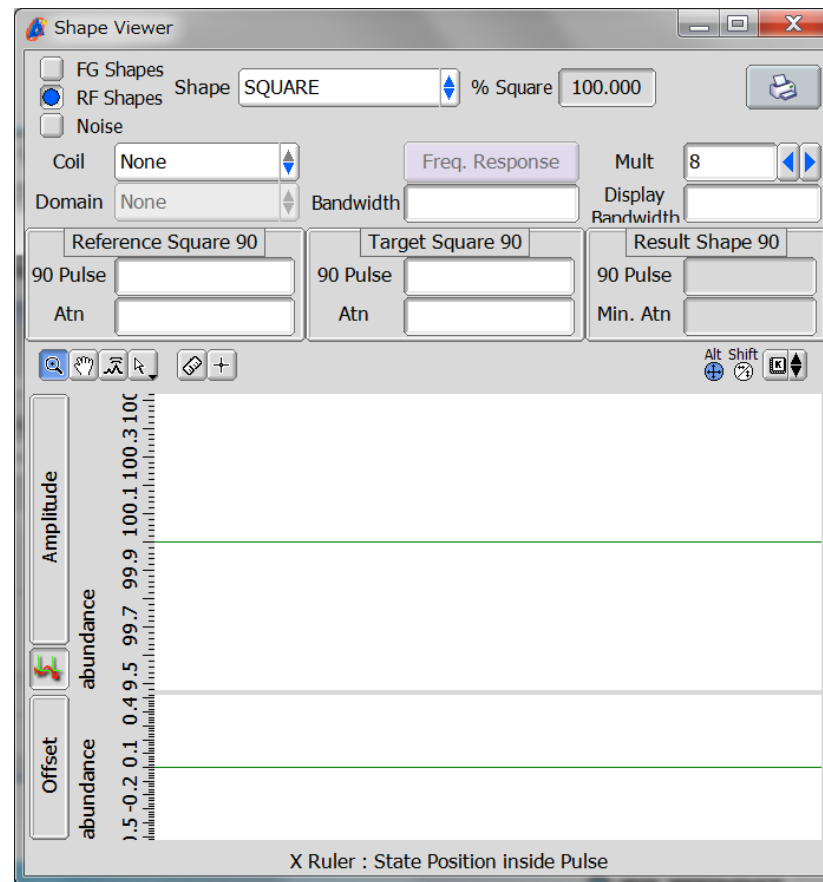


# Shape Viewer

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# Shape Viewer

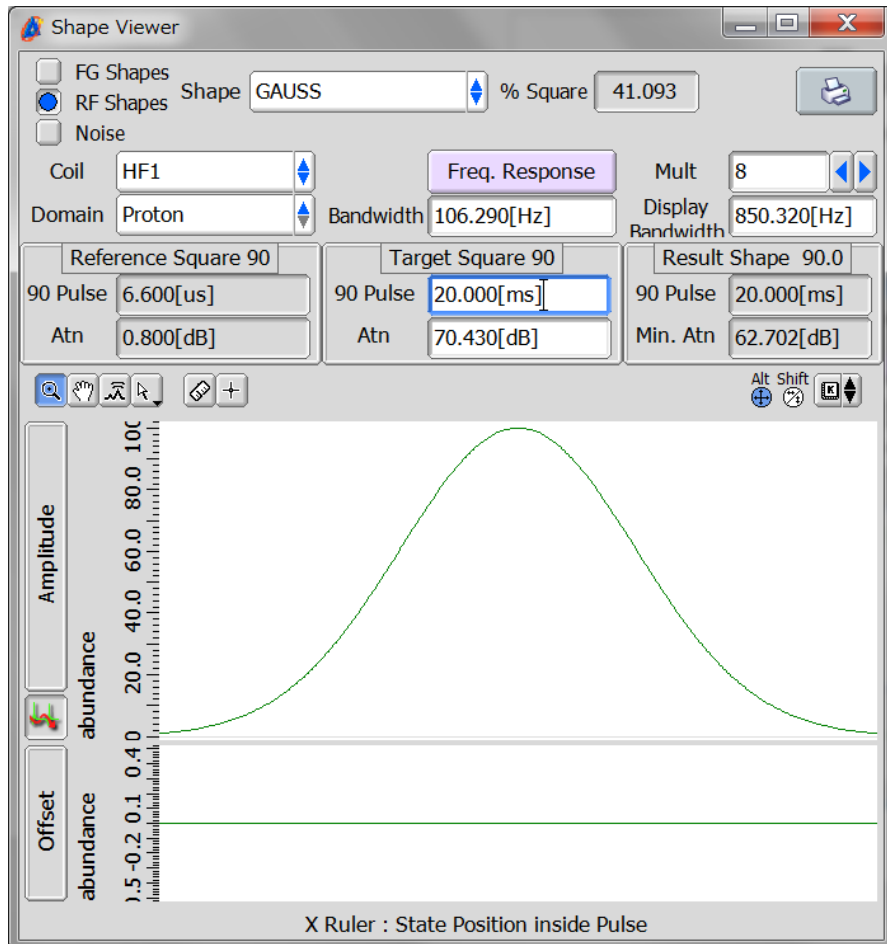
The tool which calculates the attenuator value for each shaped pulses automatically from the value of square pulse



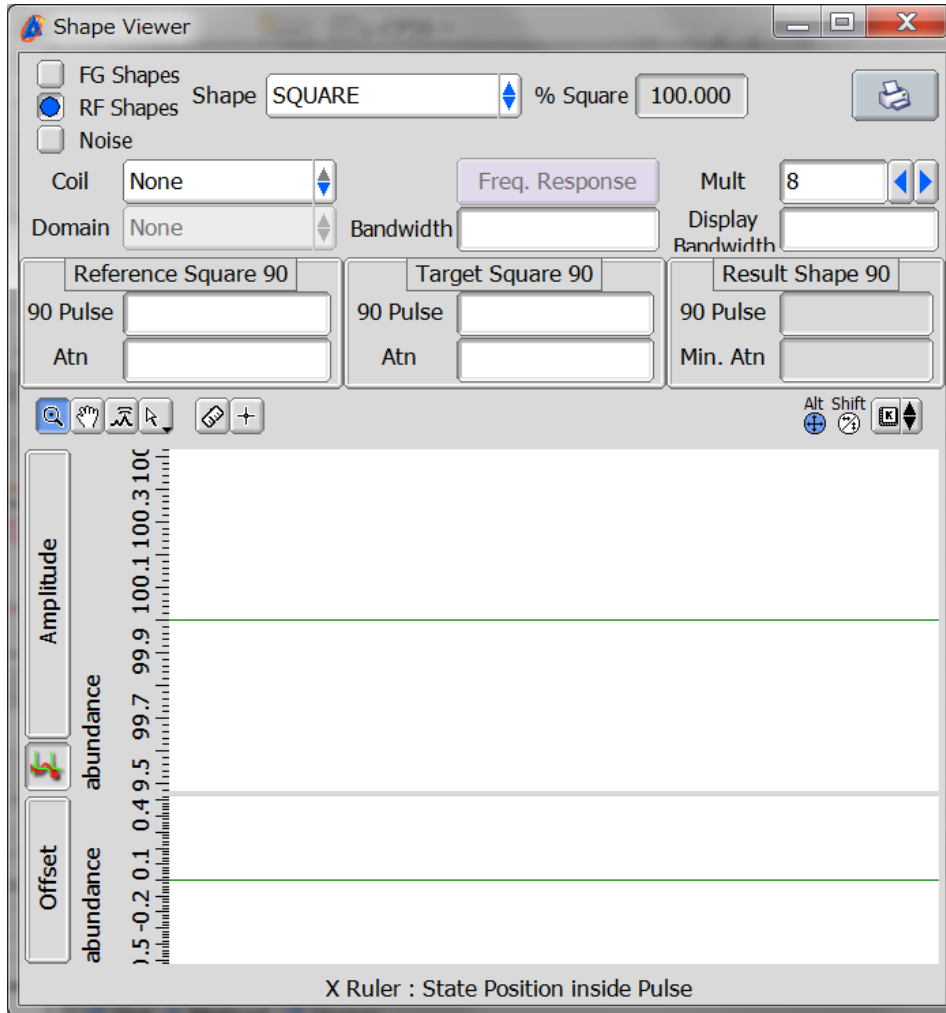
# Functions of Shape Viewer

## Main functions of Shape Viewer

- Calculation of pulse width
- Graphical shape review
- Profile of excitation
- Simulation of exciting region

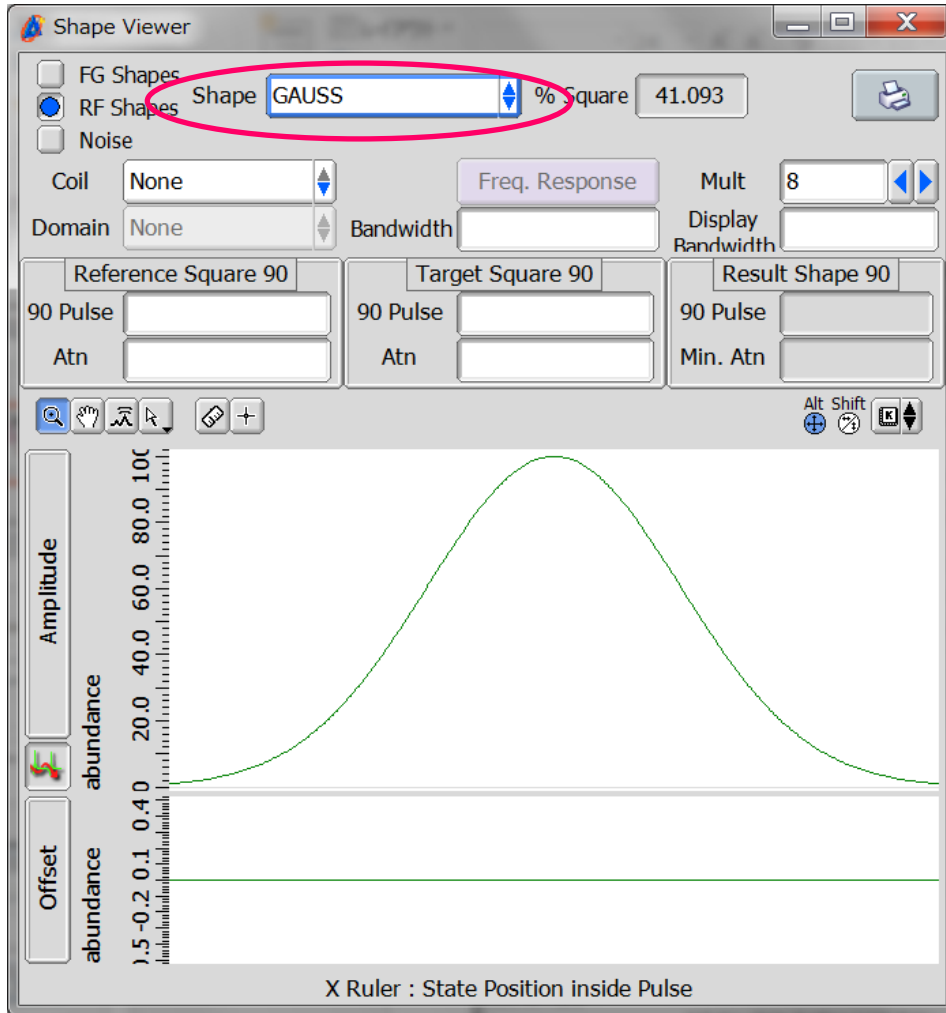


# How to calculate the pulse width



1. Select the target shape
2. Select a nucleus to “Domain”
3. Input the target pulse width

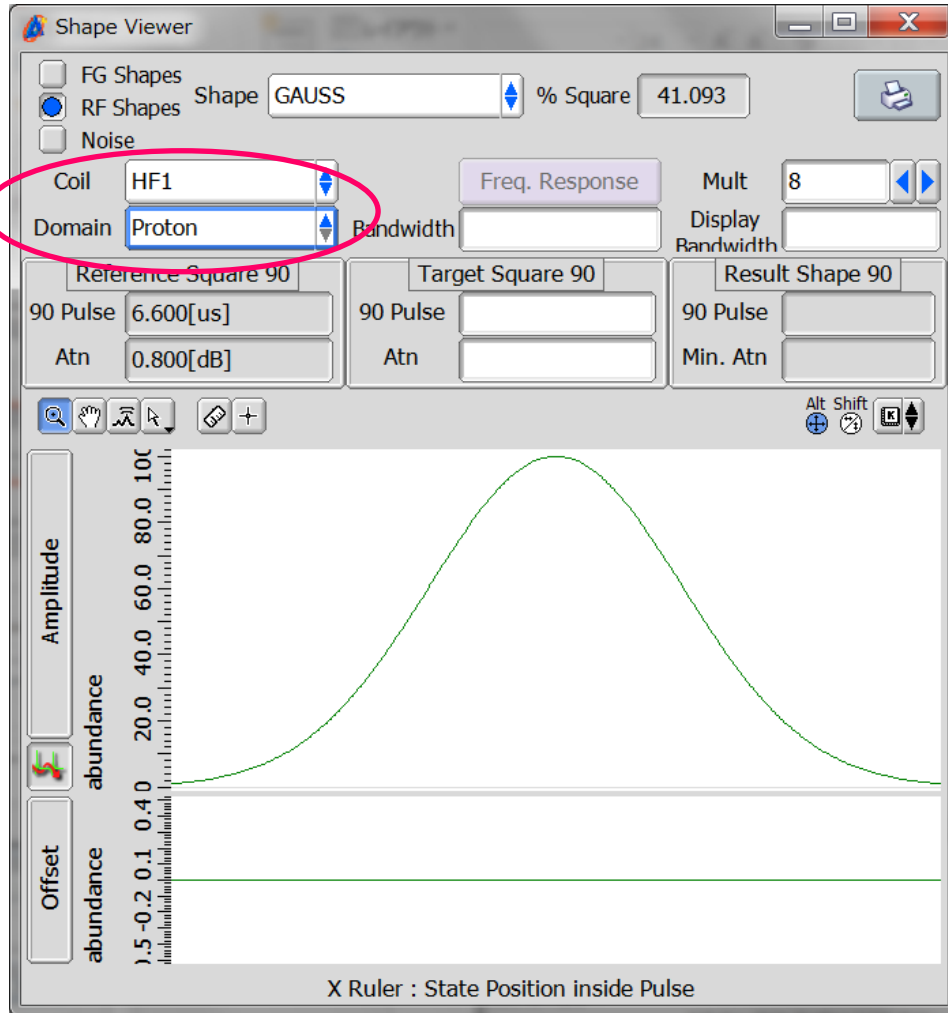
# How to calculate the pulse width



1. **Select the target shape**
2. **Select a nucleus to “Domain”**
3. **Input the target pulse width**

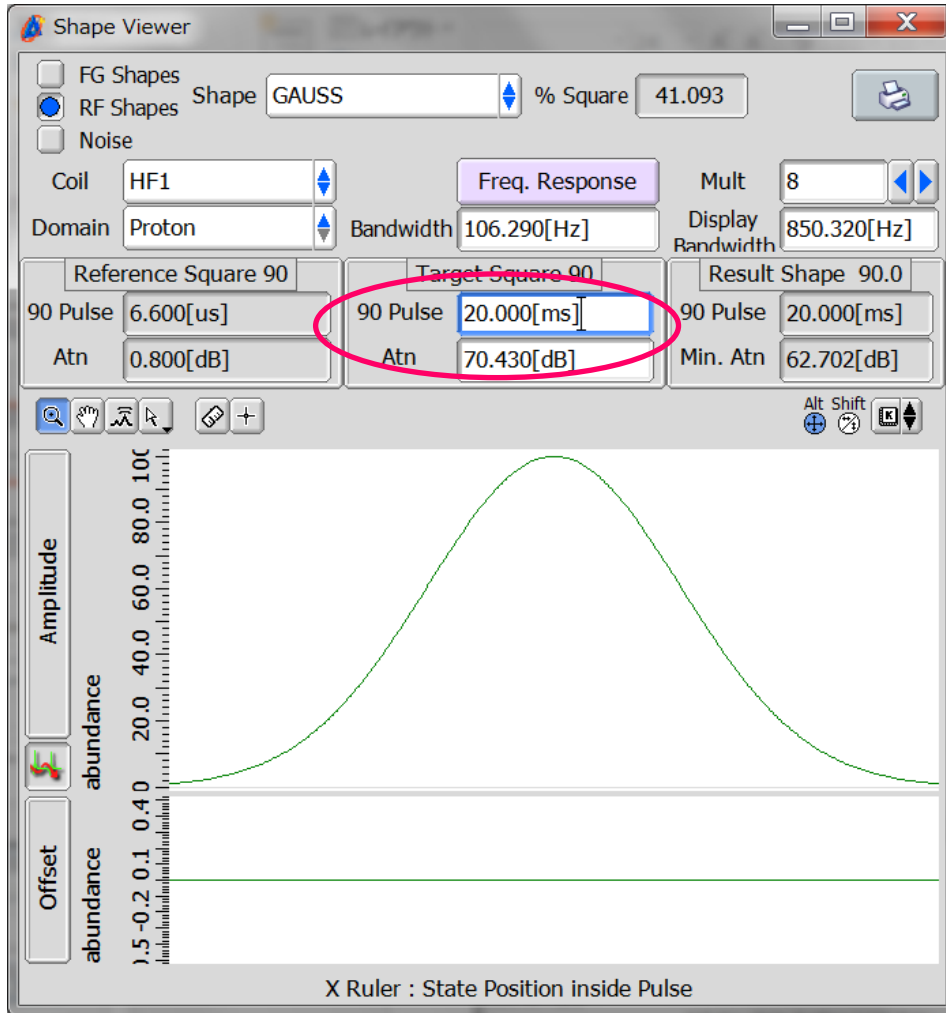


# How to calculate the pulse width



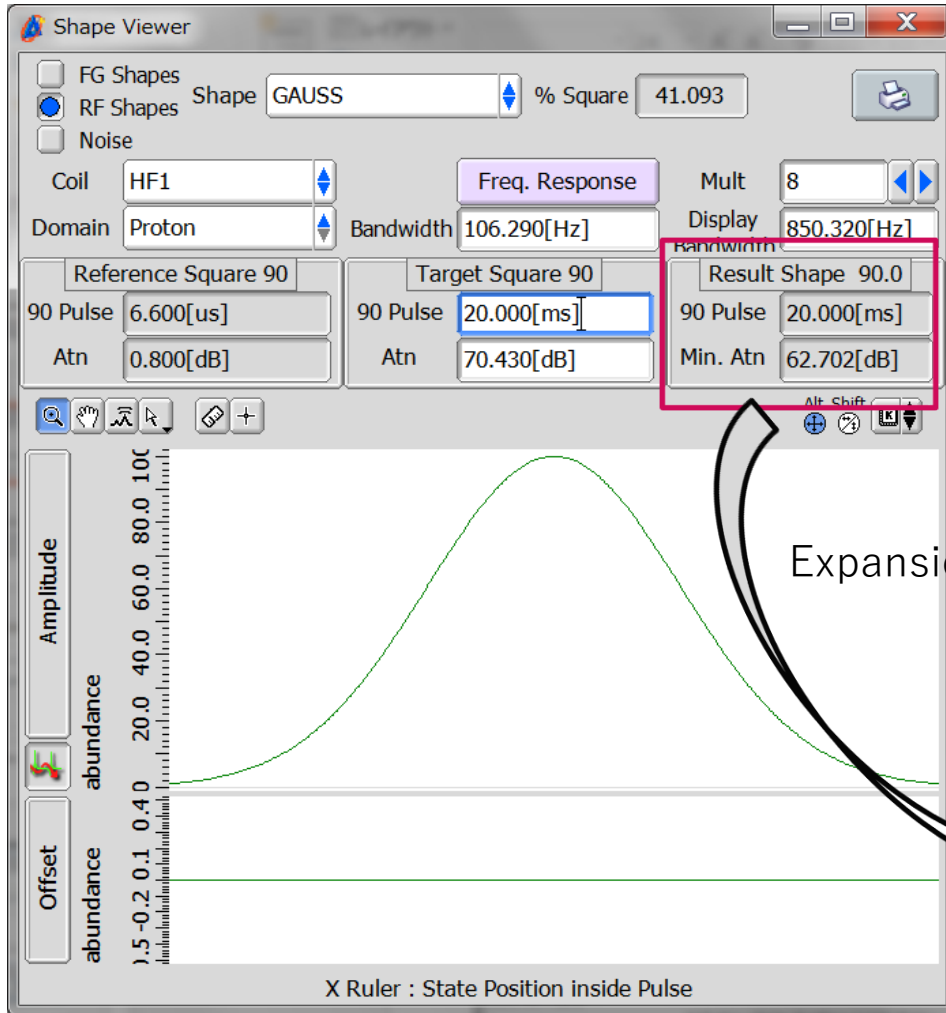
1. Select the target shape
2. Select a nucleus to “Domain”
3. Input the target pulse width

# How to calculate the pulse width



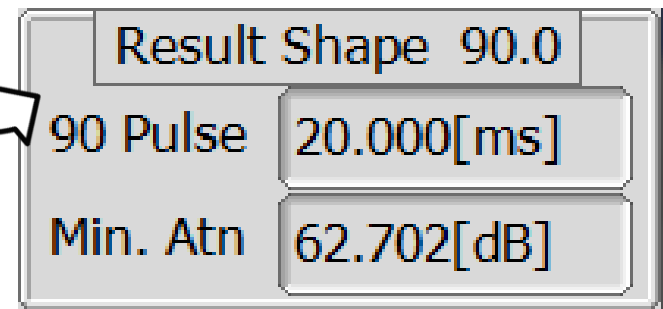
1. Select the target shape
2. Select a nucleus to “Domain”
3. Input the target pulse width

# How to calculate the pulse width



1. Select the target shape
2. Select a nucleus to “Domain”
3. Input the target pulse width

The result for GAUSS pulse



# How to put the result into experiment

Pulse width of selective reverse  
(180 degree) pulse

obs\_sel\_180 40[ms] x90\_soft \* 2

Attenuator of selective pulse

obs\_sel\_shape GAUSS

Offset of selective pulse

obs\_sel\_atn 70.43[dB] soft\_atn\_calc

Shape of selective pulse

obs\_sel\_offset 5[ppm] x\_offset

soft\_atn\_calc 70.43[dB]

... from V5.1.0

Automatically calculated

obs\_sel\_180 40[ms] x90\_soft \* 2

obs\_sel\_shape GAUSS

soft\_bandwidth\_hz 53.15[Hz]

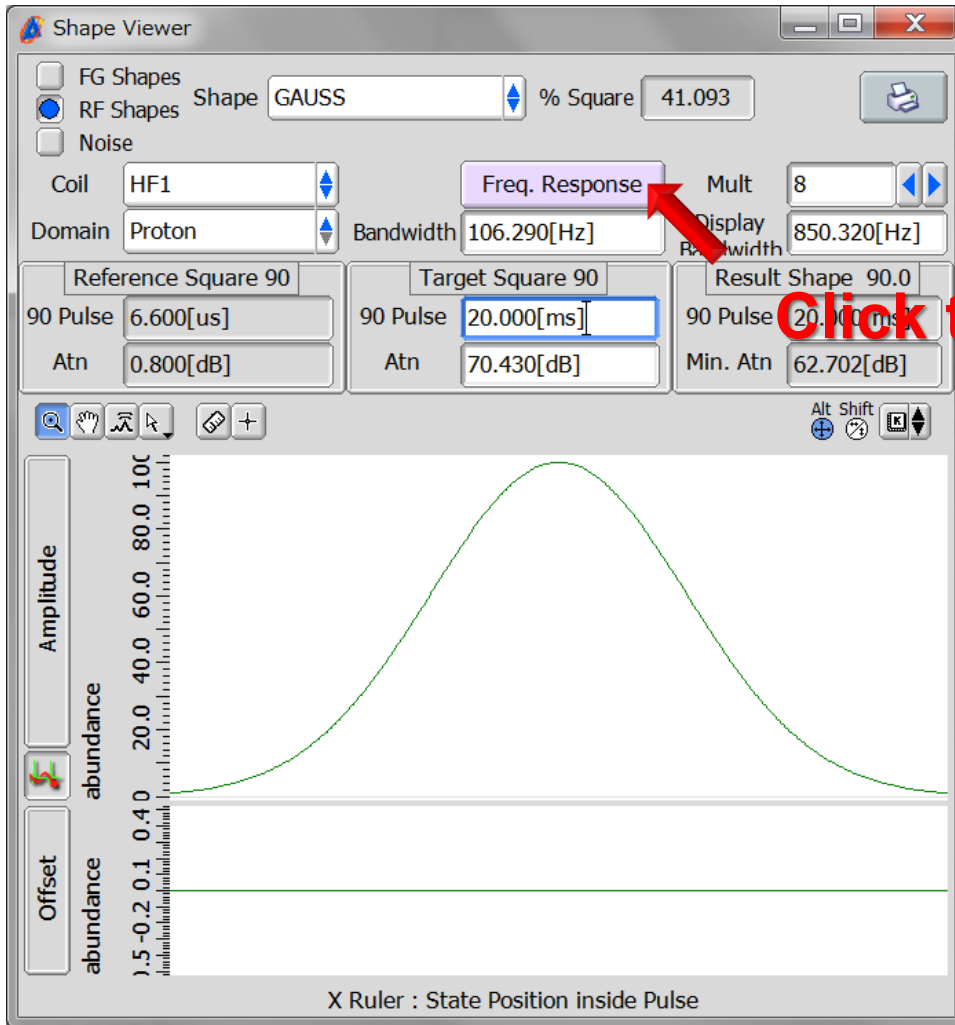
soft\_bandwidth\_ppm 0.13295[ppm]

soft\_atn\_calc 70.43[dB]

obs\_sel\_atn 70.43[dB] soft\_atn\_calc

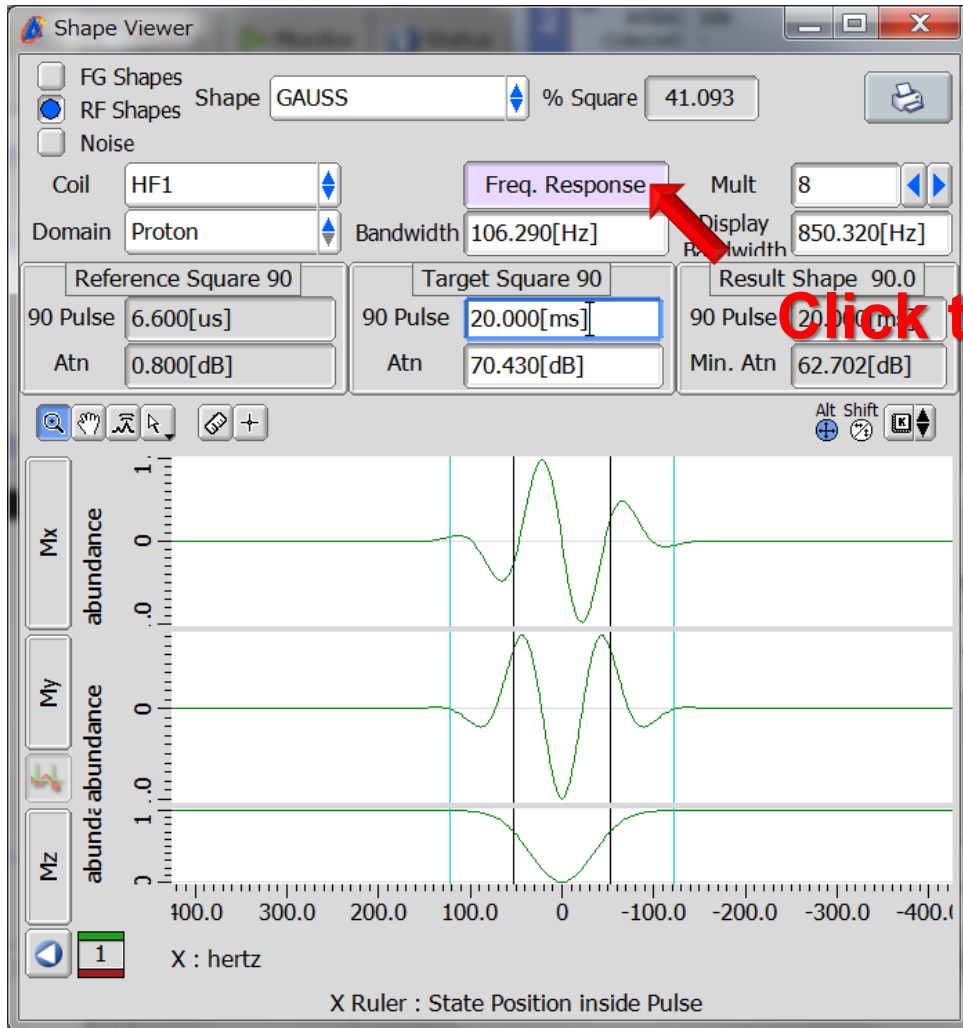
obs\_sel\_offset 5[ppm] x\_offset

# Display the profile of Frequency Response



Click the "Freq. Response" button

# Display the profile of Frequency Response

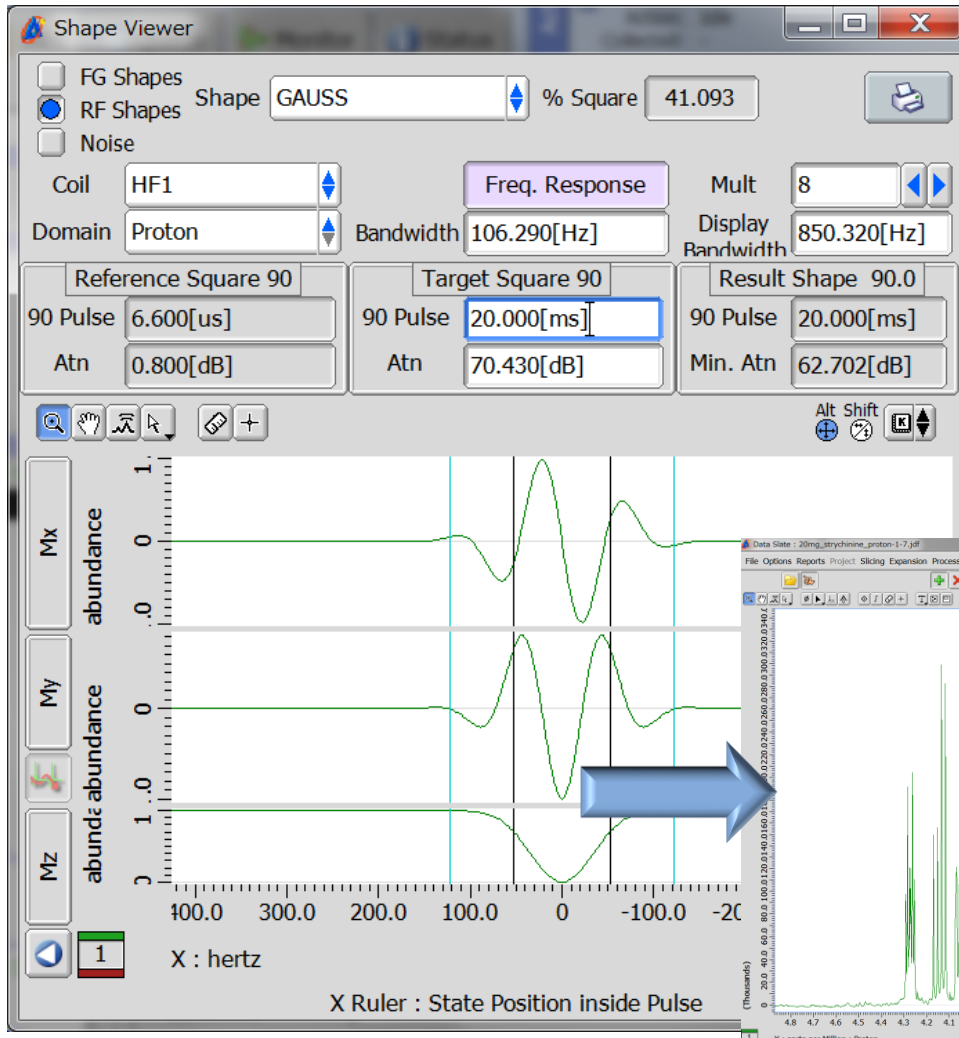


Click the "Freq. Response" button

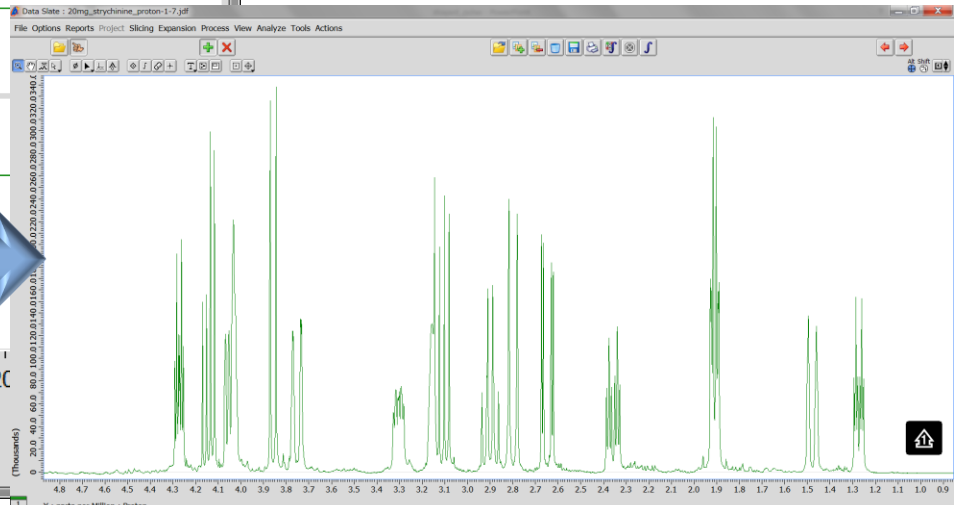
The excitation profile for X axis and Y axis

The excitation profile for Z axis

# Compare the profile with the actual spectrum



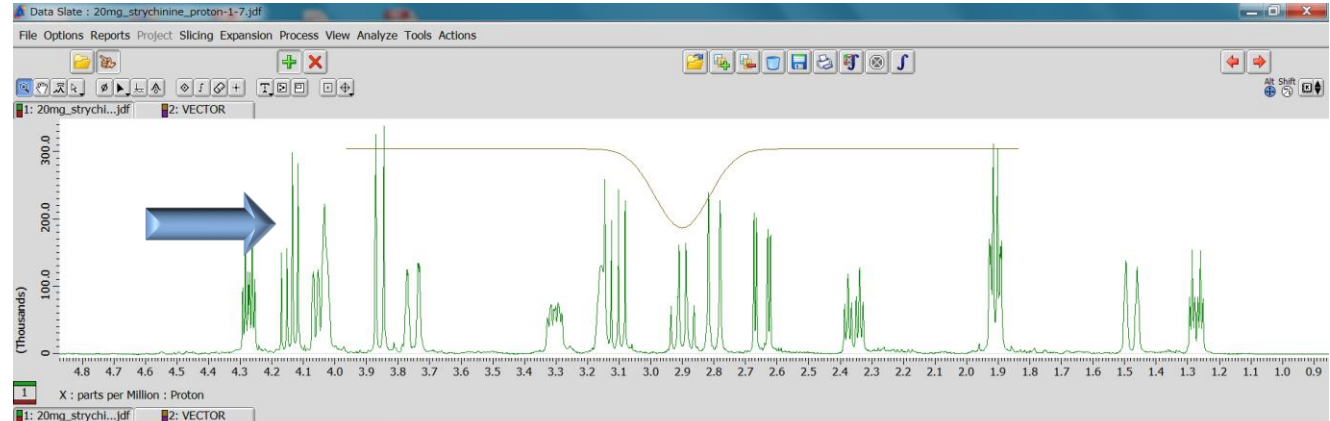
**Note: Remove the ppm step from the processing list of the dataset before exporting it to data slate in order to overlay**



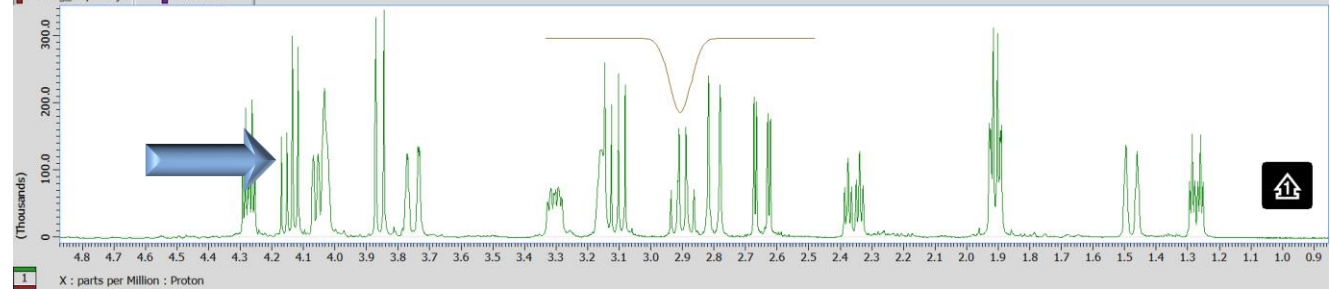
**Overlay the profile to spectrum**

# Compare the profile with the actual spectrum

20[ms] Gaussian pulse



50[ms] Gaussian pulse



50[ms] is better for selecting the peak at 2.9[ppm]

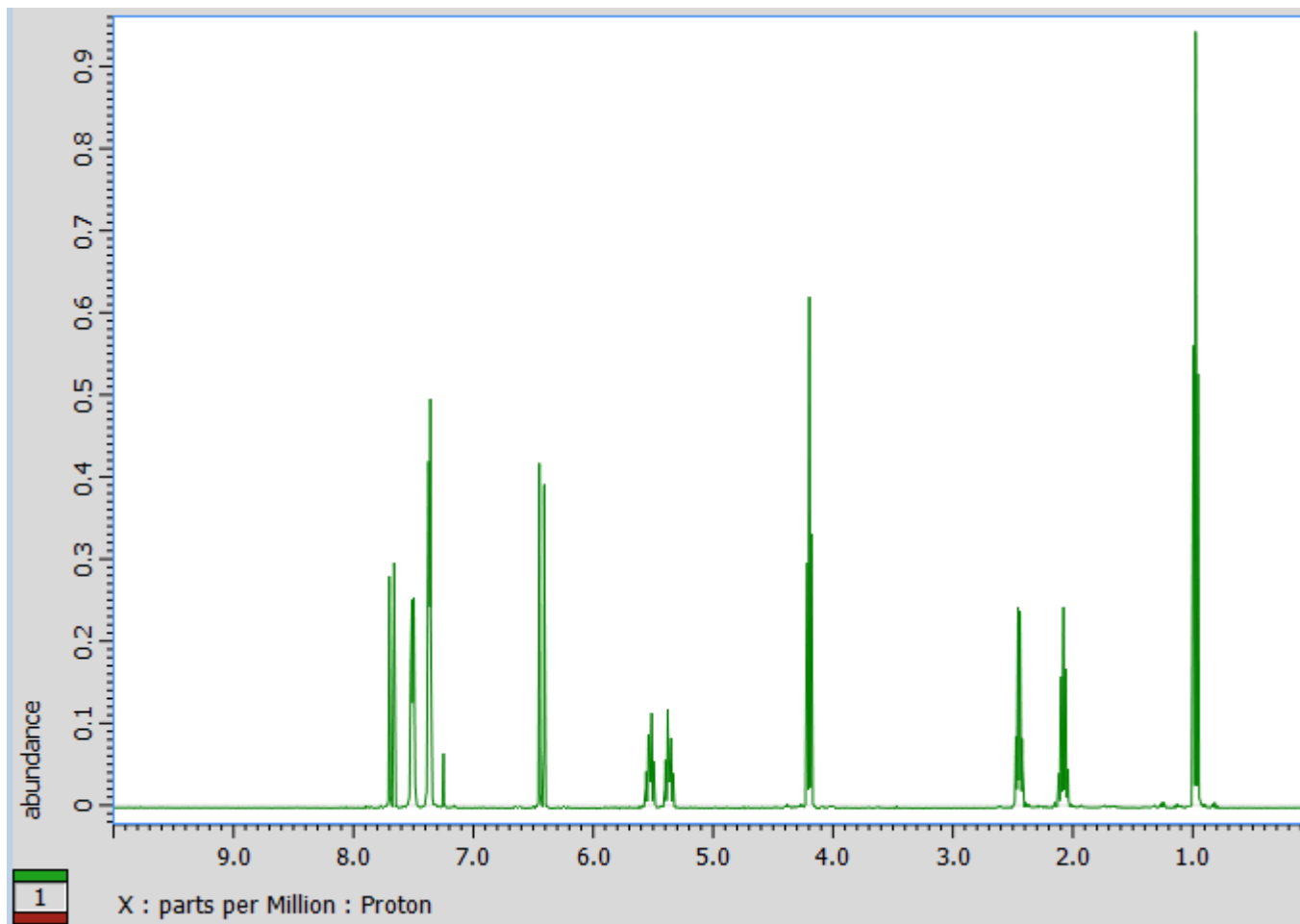


# Pure shift

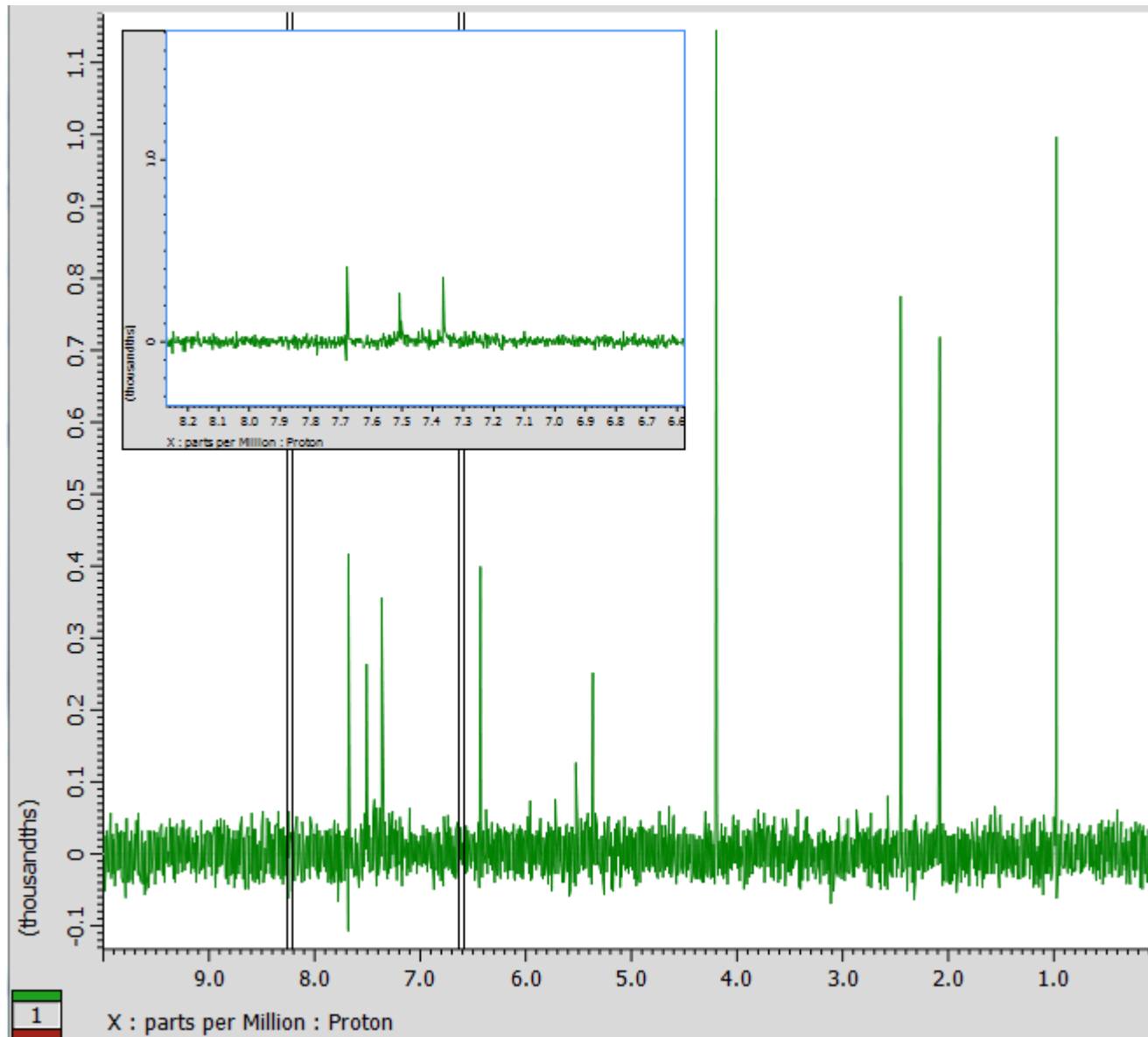
---

<https://www.nmr.chemistry.manchester.ac.uk/?q=node/421>

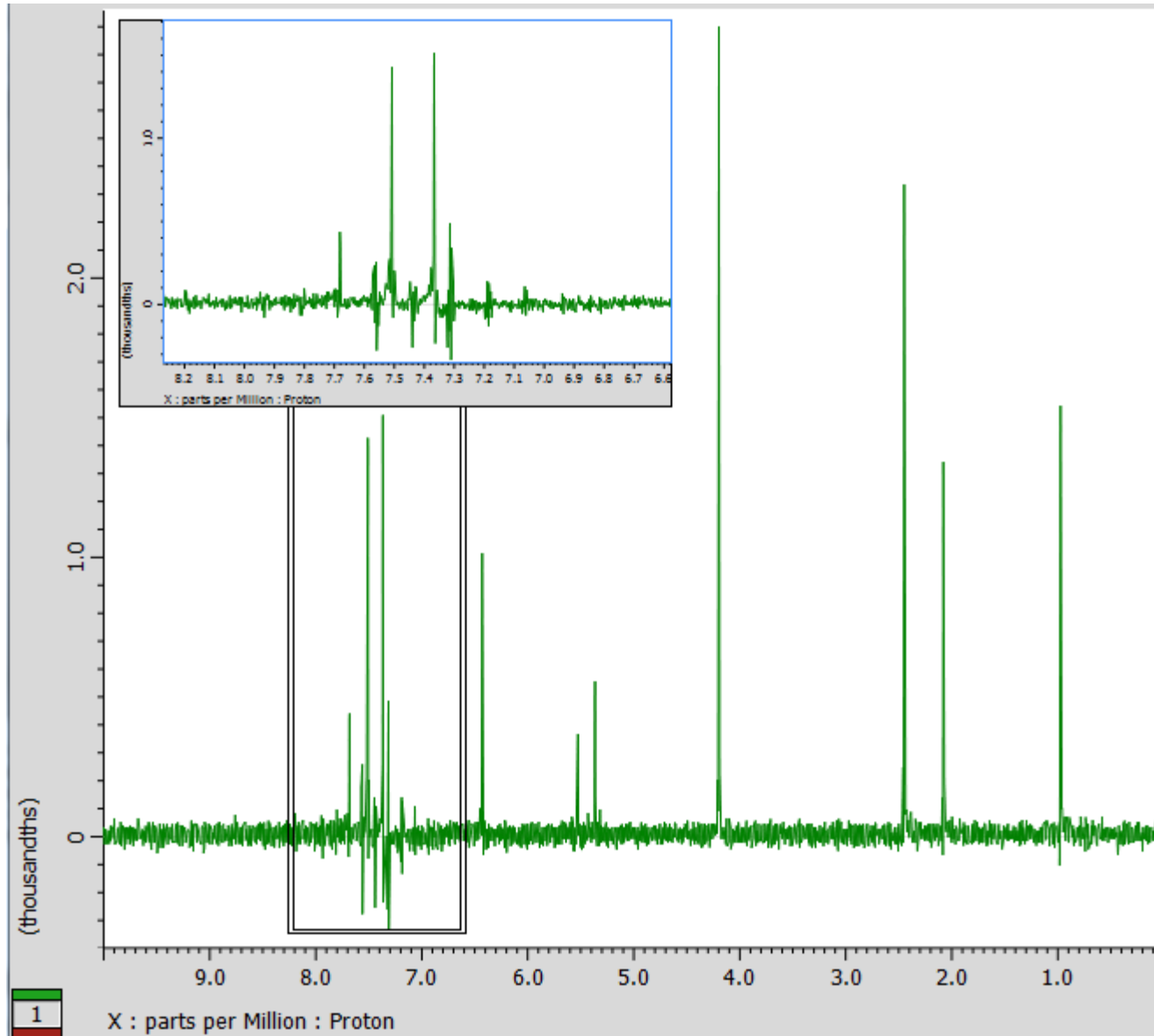
# Proton



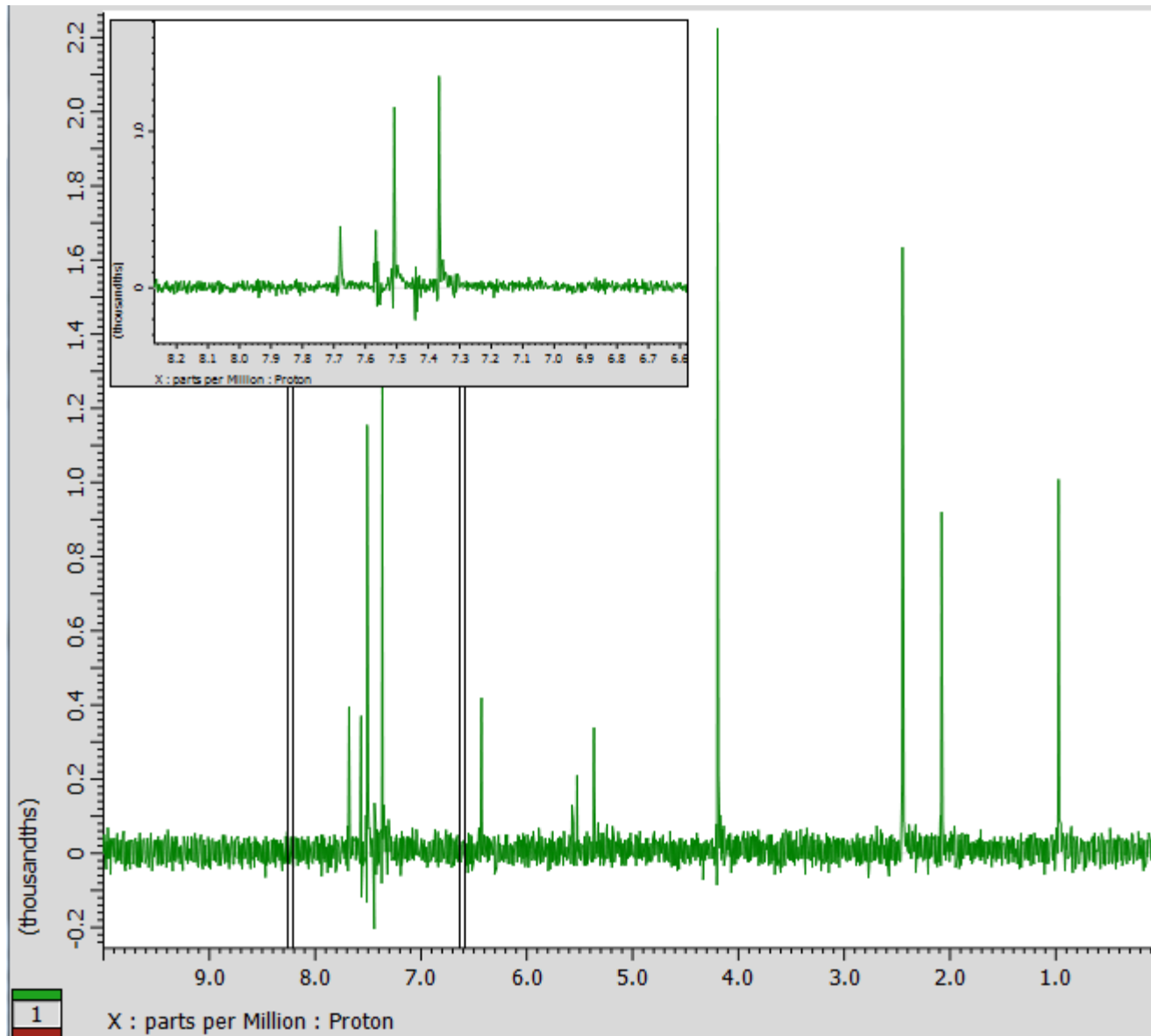
# Zangger-Sterk



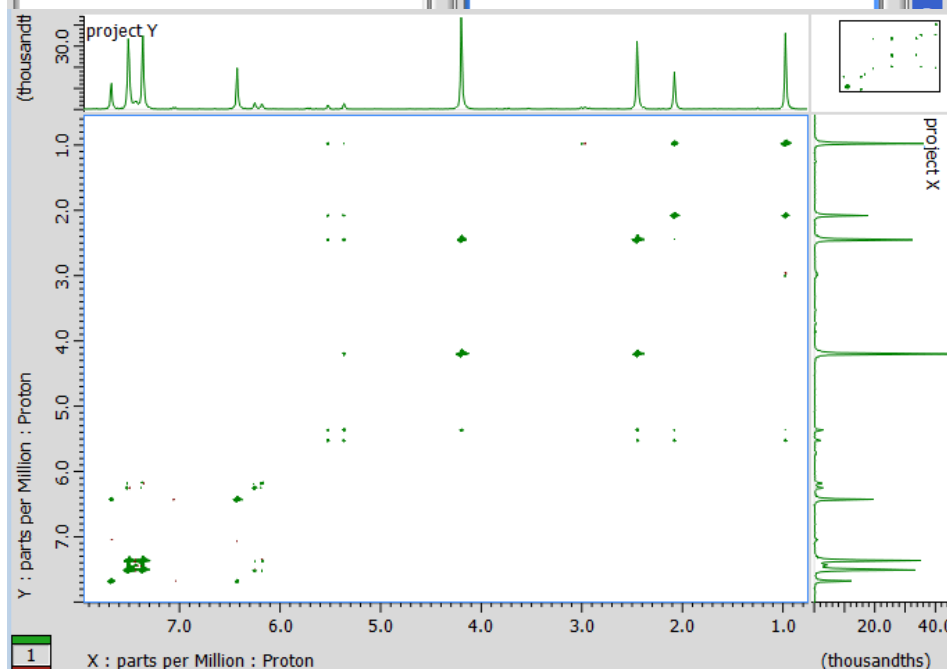
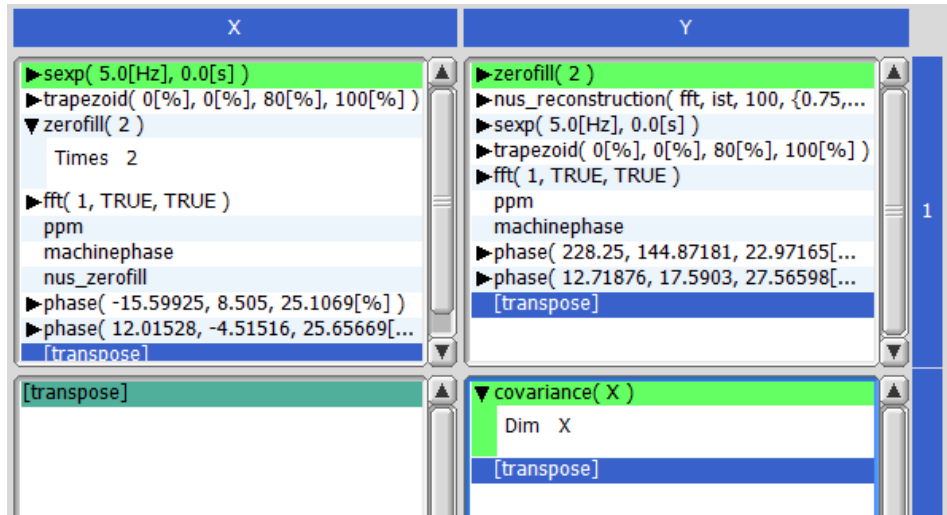
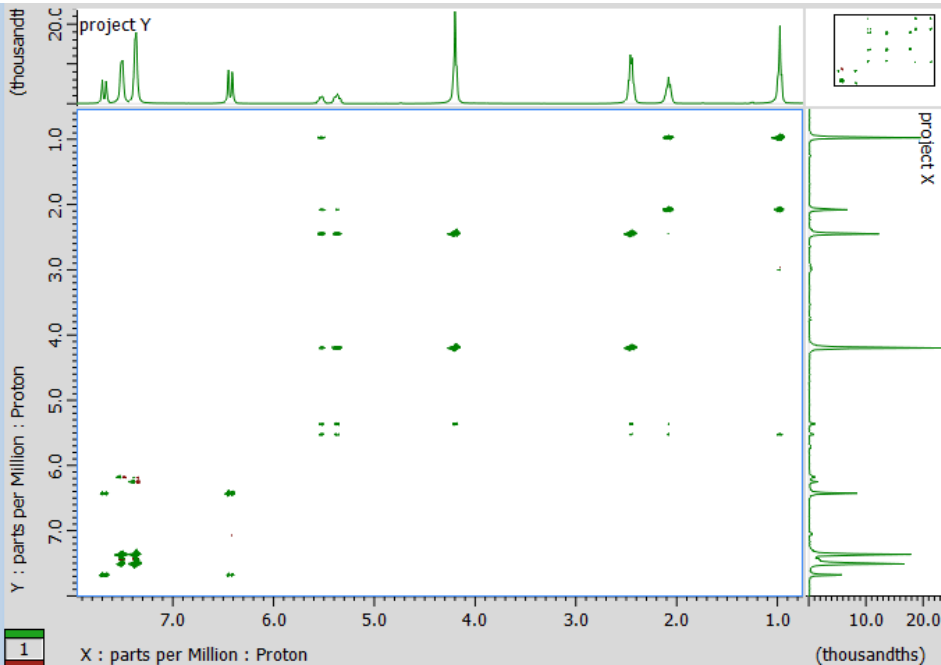
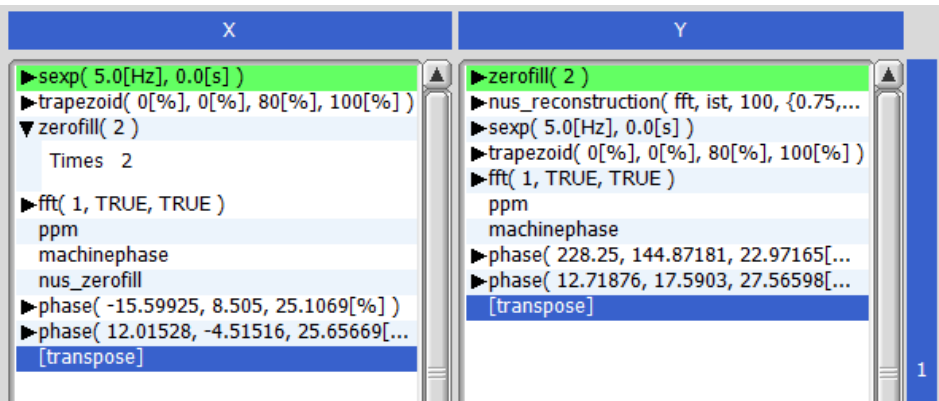
# PSYCHE



# TSE-PSYCHE



# PSYCHE-TOCSY (F1) + NUS + Covariance



# Relaxation

---

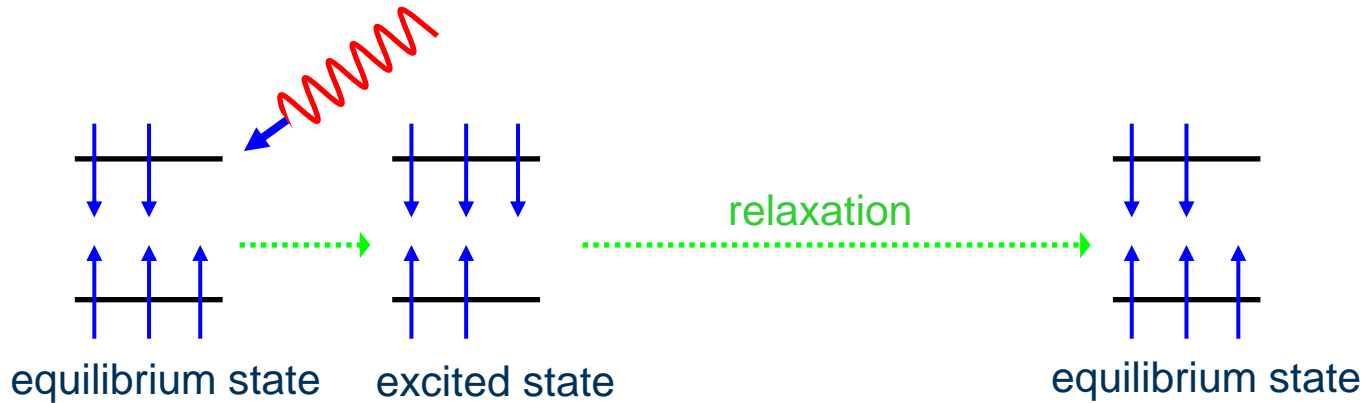
# Relaxation

---

- Exponential processes
- Longitudinal, or spin-lattice, relaxation ( $T_1$ )
- $T_1$  relaxation mechanisms
- Transverse, or spin-spin, relaxation ( $T_2$ )

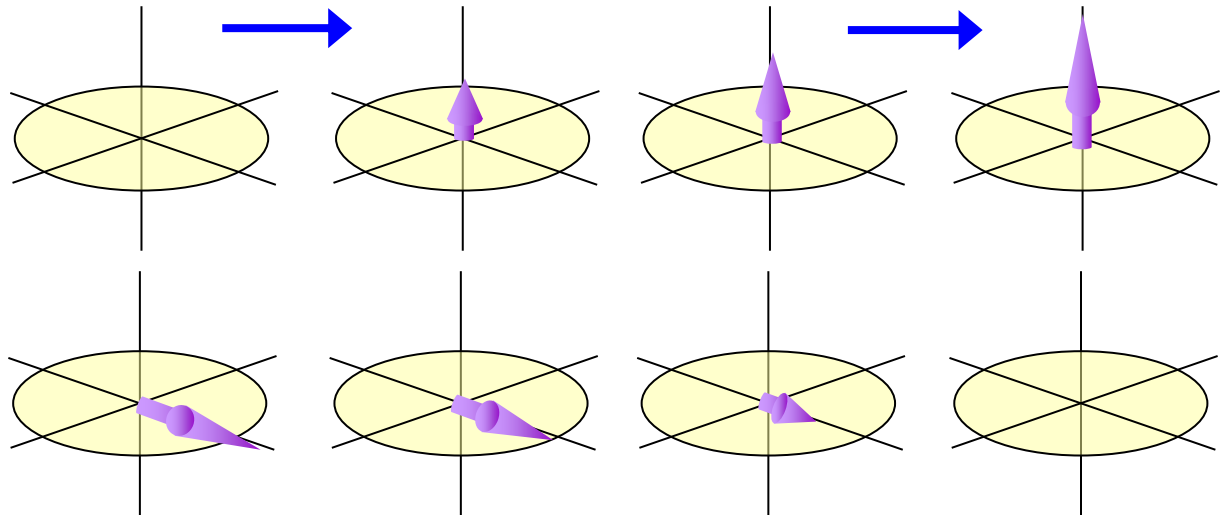


# Relaxation



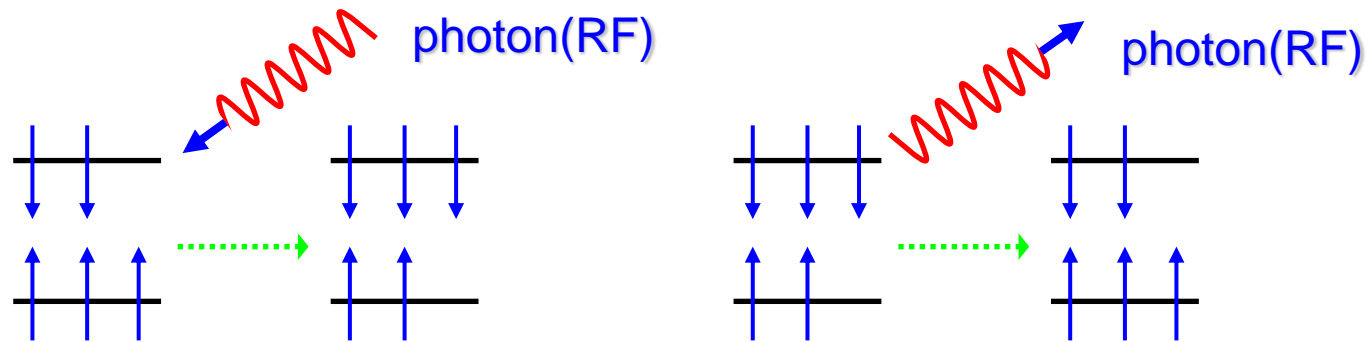
Z component recovers to the initial equilibrium state

$T_1$  relaxation



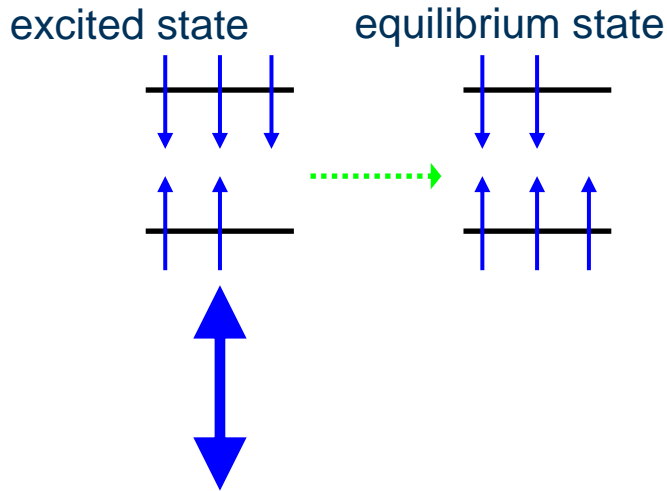
XY component decays to zero (also the initial equilibrium state)

# What causes relaxation in NMR?



In case UV or VIS, the energy gap  $\Delta E$  is large and the relaxation is dominated by spontaneous emission of photon. However, the energy gap in NMR is extremely small. If spontaneous emission of photon dominates the relaxation in NMR,  $T_1$  is estimated to  $10^9$ - $10^{10}$ sec (30years-300years!) even at a high field machine

# What causes relaxation in NMR?



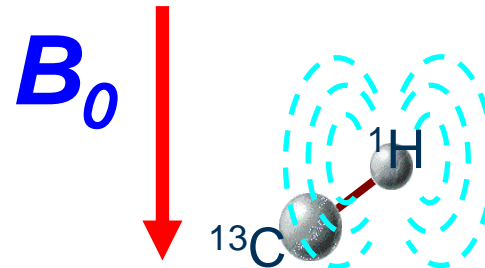
NMR relaxation is the random NMR transition caused by interaction with thermal reservoir (or lattice)

**Thermal reservoir**

Any time dependent interaction can be "thermal reservoir", which are molecular rotation, vibration, collisions ,,,,

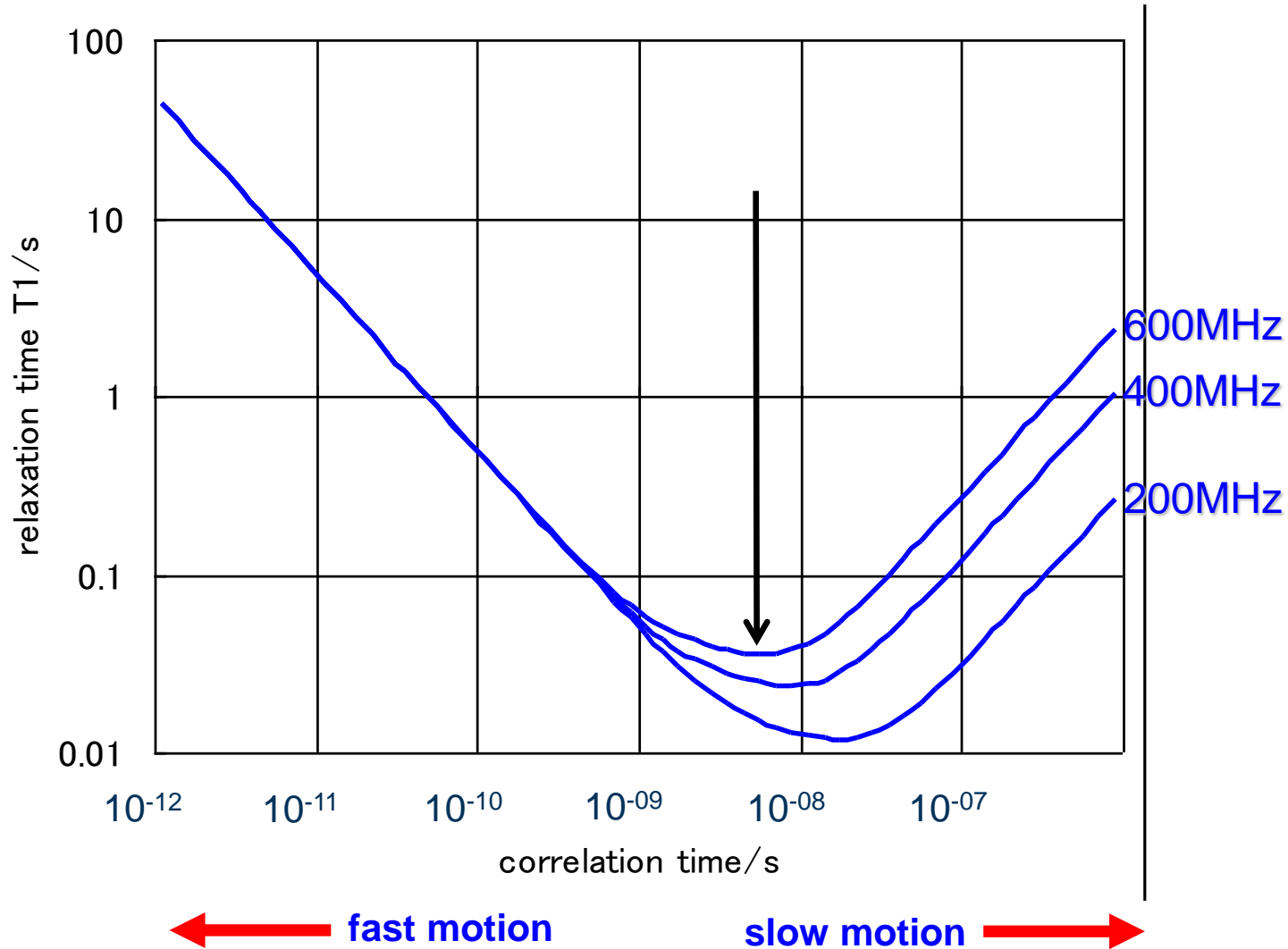
The primary interaction to cause NMR relaxation in liquid state is dipole-dipole interaction between two nuclear spins.

$^{13}\text{C}$  at the right picture experiences dipole field from  $^1\text{H}$ . The dipole field changes an angle to  $B_0$  with molecular tumbling in liquid.



# Molecular motion vs $T_1$

Molecular motion = resonance frequency at a 600MHz machine



# T<sub>1</sub> mechanisms

- Relaxation occurs when nearby local fields fluctuate at the Larmor frequency of a nucleus
  - Chemical shift anisotropy (CSA)
  - Quadrupolar interactions (Q)
  - Scalar coupling (SC)
  - Electron-spin interactions (ES)
  - Spin rotation (SR)
  - Dipole-dipole interactions (DD)

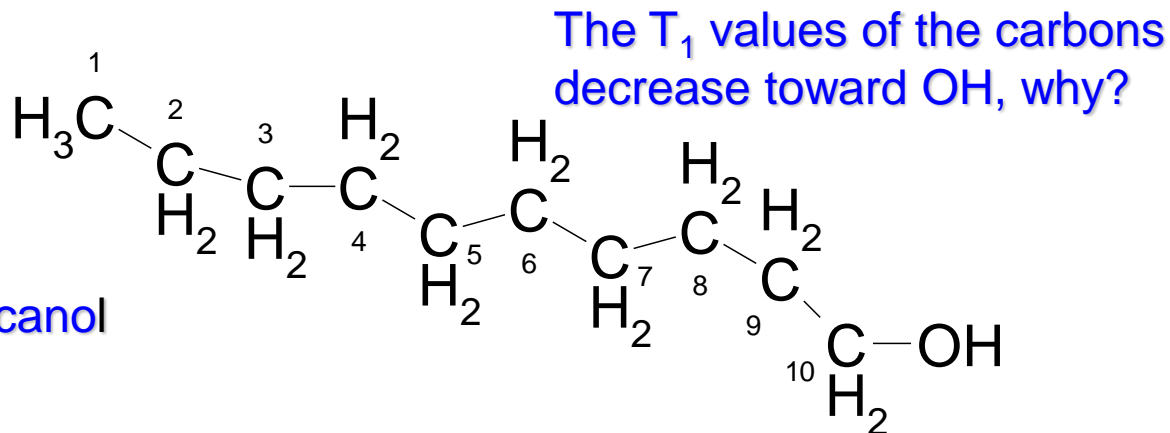
# Relaxation related Information and NMR parameter

---

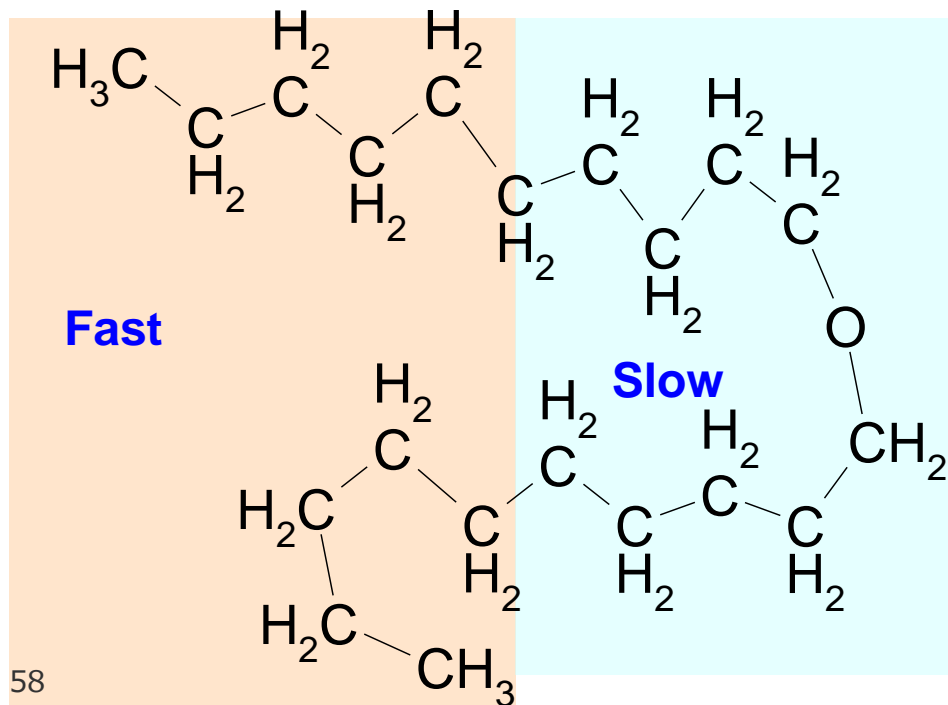
- Molecular structure
- Molecular mobility
- Inter-molecule interaction
- Chemical exchange
- NMR spectrum and experimental condition
  - sensitivity
  - accuracy of integration
  - pulse program parameter optimization

# Molecular interaction elucidated by NMR relaxation

n-decanol

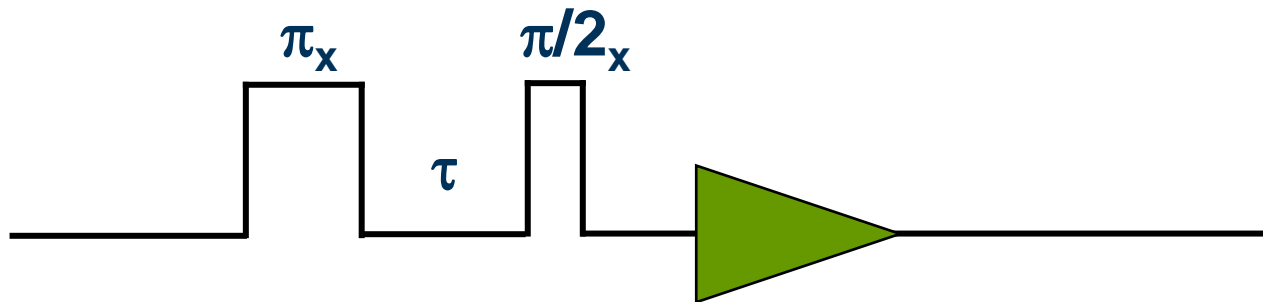


Carbon	$T_1$ /sec
1	3.1
2	2.2
3	1.6
4	1.1
5	0.84
6	0.84
7	0.84
8	0.77
9	0.77
10	0.65



Those  $T_1$  values prove the two molecular complex by hydrogen bound

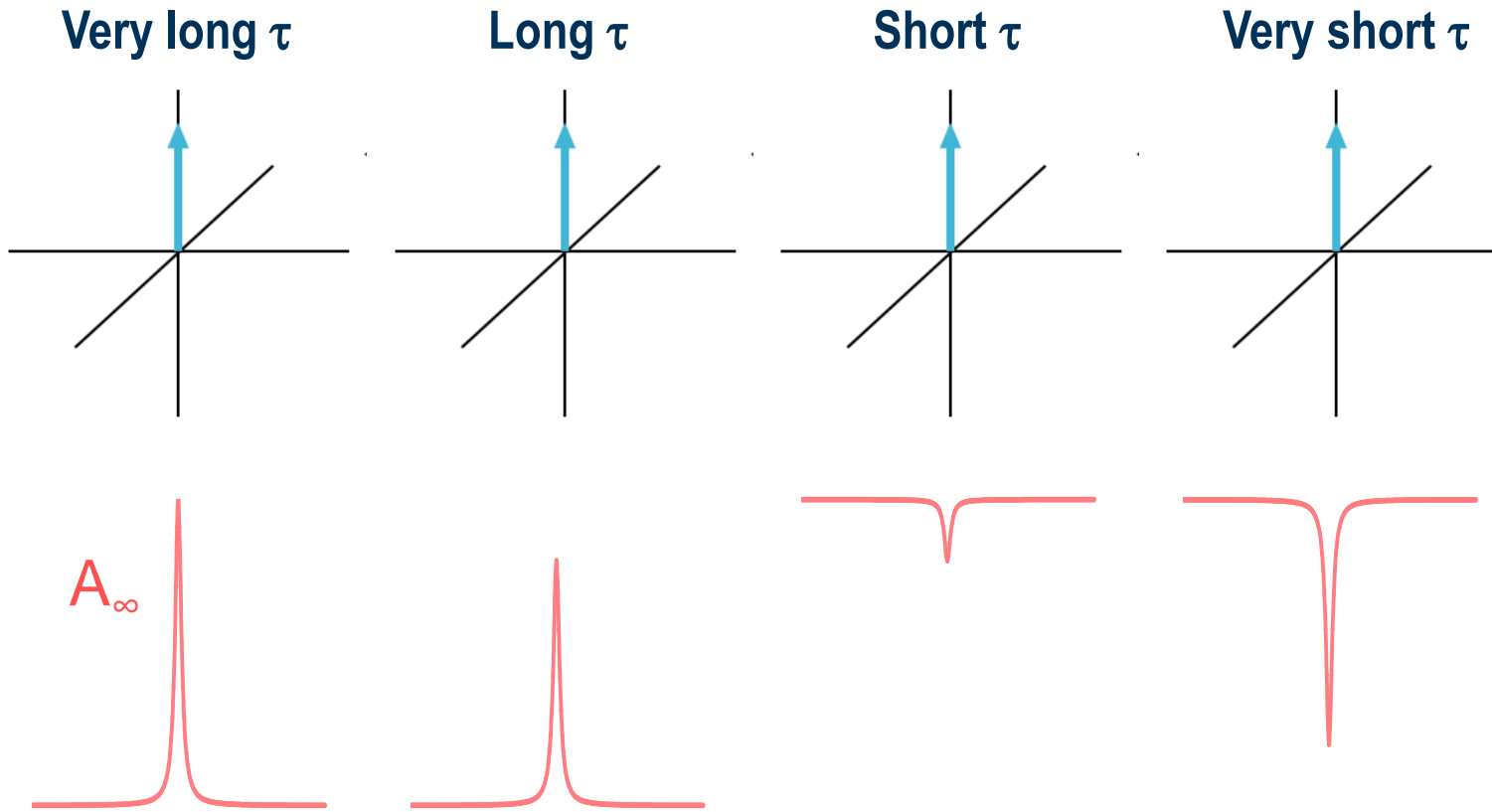
# $T_1$ by inversion recovery



- Measure intensity as  $\tau$  is varied
- Plot  $\ln (A_\infty - A)$  vs.  $\tau$  or fit data

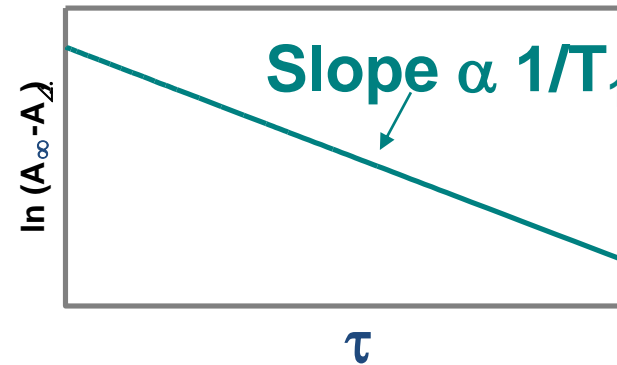
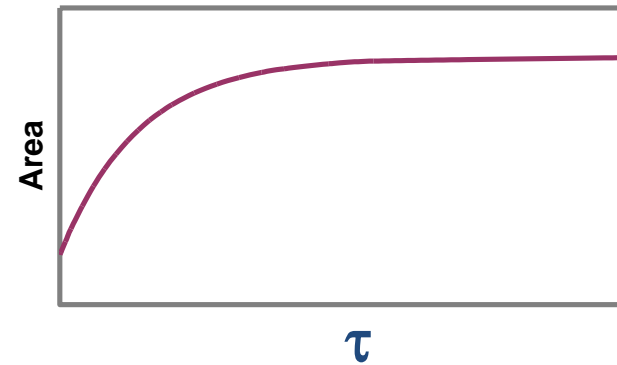
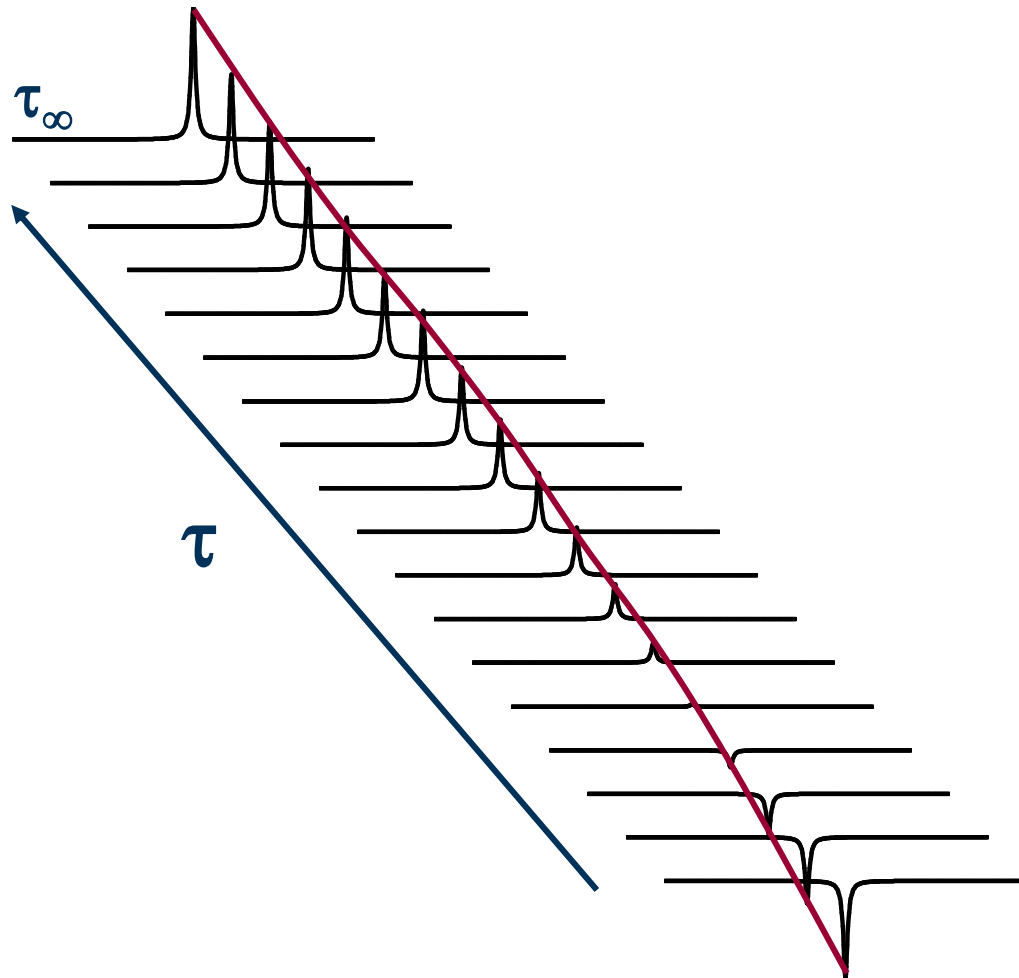


# $T_1$ by inversion recovery

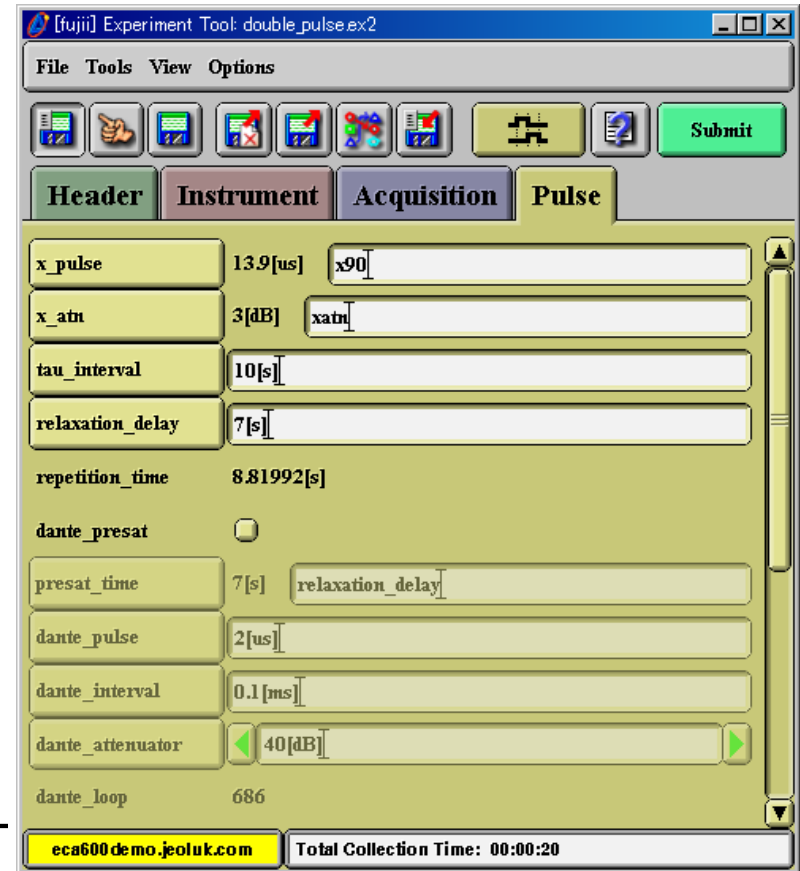
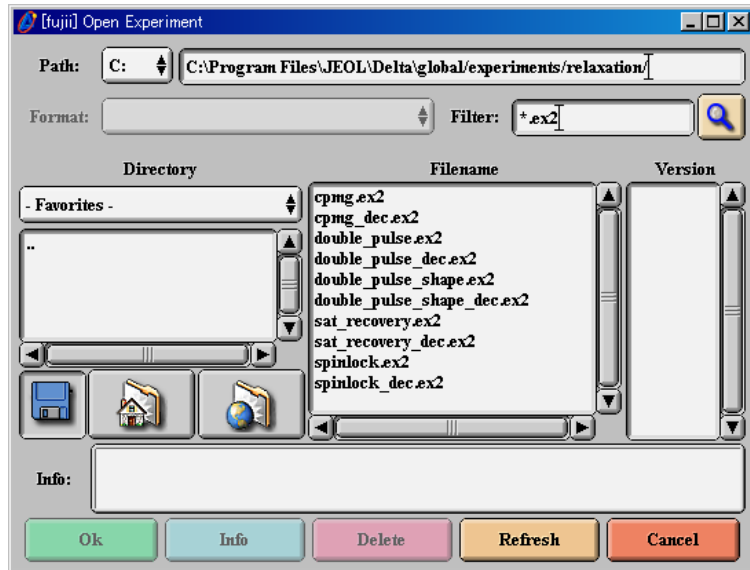


$$A = A_\infty (1 - 2e^{-\tau/T_1})$$

# Area versus $\tau$



# “Quick” $T_1$ test



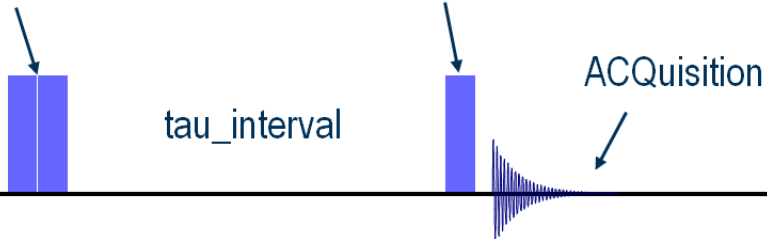
180° pulses( $x\_pulse*2$ )

90° pulses( $x\_pulse$ )

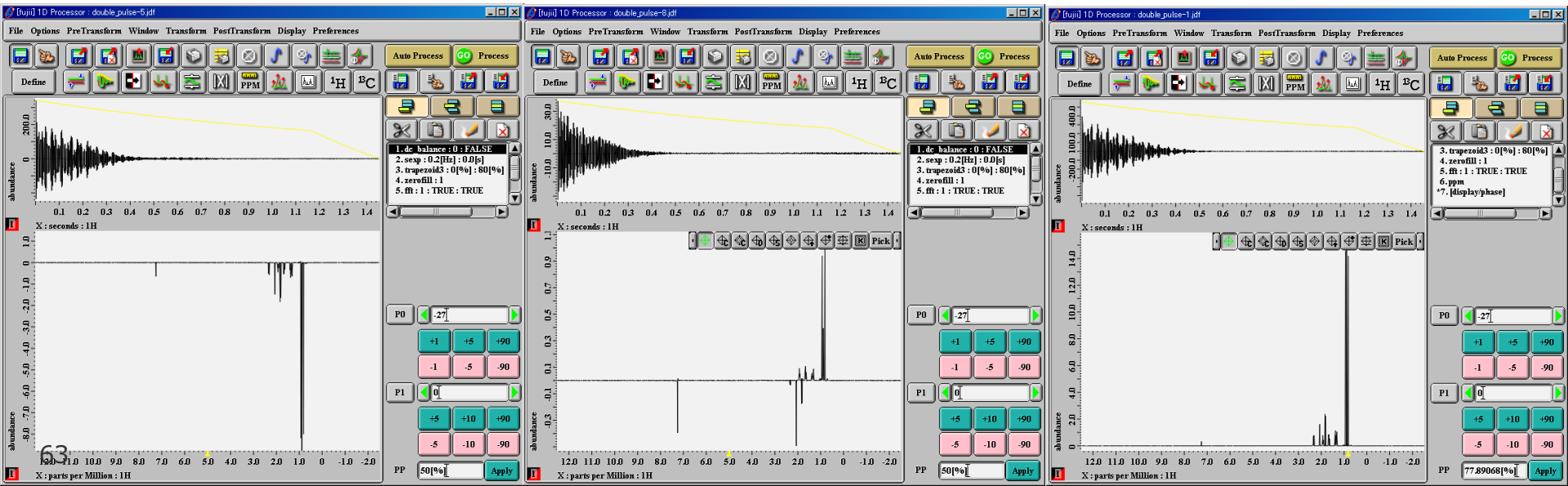
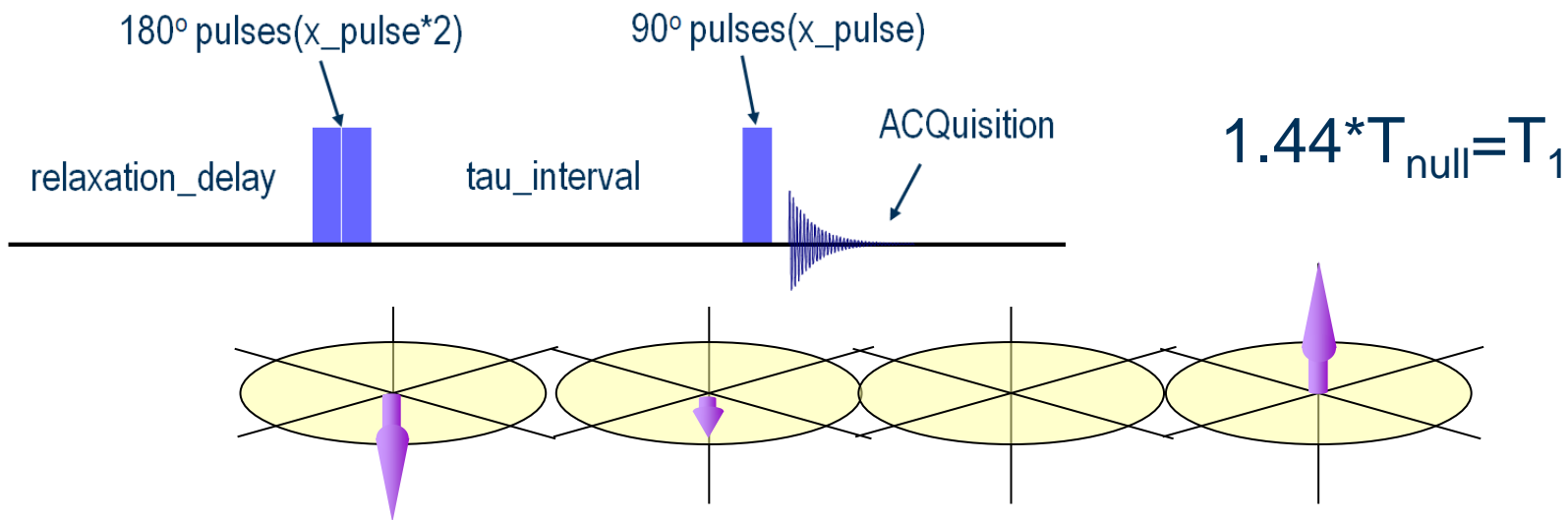
relaxation\_delay

tau\_interval

ACQuisition

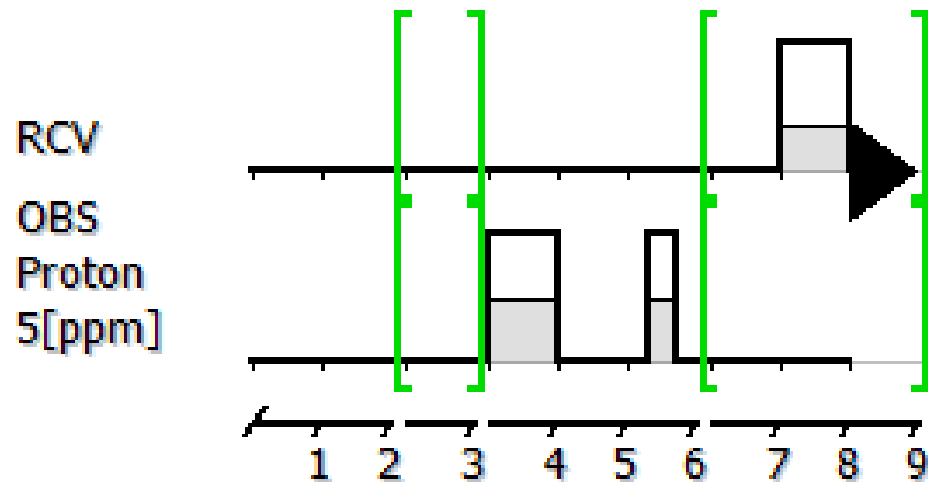
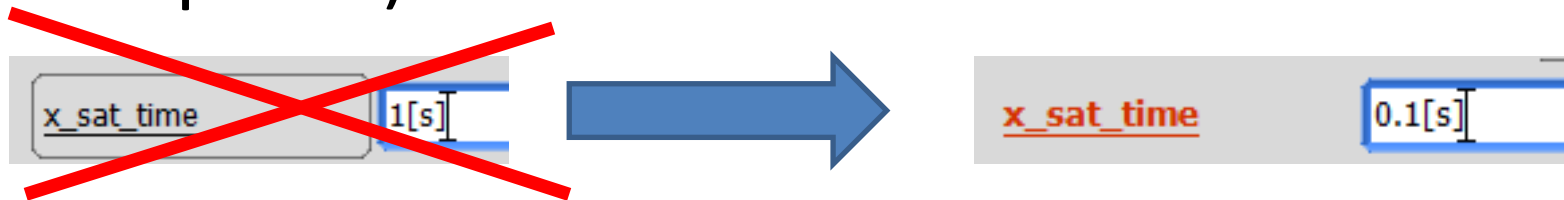


# “Quick” $T_1$ test



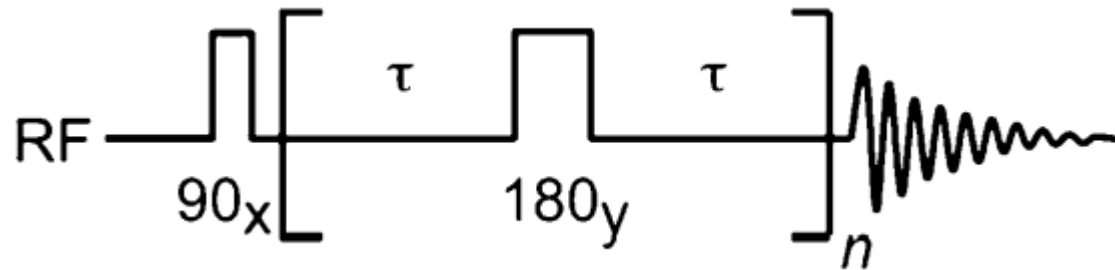
# Quick $T_1$ measurement: Saturation recovery

- Avoid long recovery delays relaxation\_delay 1[s]
- Correct the saturation time (1[s] may damage the probe)



# T<sub>2</sub> measurement

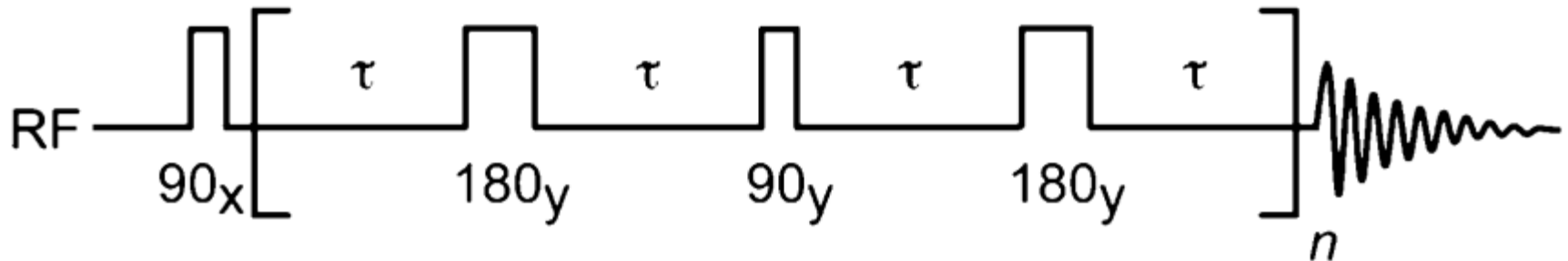
- CPMG



- Increase  $n$  to increase the time magnetization is in transverse plane
- J-modulation problems
  - $\tau$  needs to be very short  $< 2\text{ms}$
  - Heating problems
  - Result is a mixture of  $T_2$  and  $T_{1\rho}$
- Isotropic mixing

# T<sub>2</sub> measurement

- Project



- Train of perfect spin echos
  - J-modulation is quenched for  $\tau J \ll 1$
- $\tau$  can be as long as 5ms
- Planar mixing (about half as fast as isotropic mixing)

# How to array tau\_interval / delay\_list?

- Saturation recovery: Exponential from  $T_1/10$  to  $10 * T_1$
- Inversion recovery: Logarithmic base 0.3 from  $T_1/20$  to  $5 * T_1$   
(if last tau\_interval is not long enough, use non-linear fitting)
- CPMG / Project: Logarithmic base 0.3 from  $T_1/50$  to  $2.5 * T_1$



# Solvent suppression

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# Solvent Suppression

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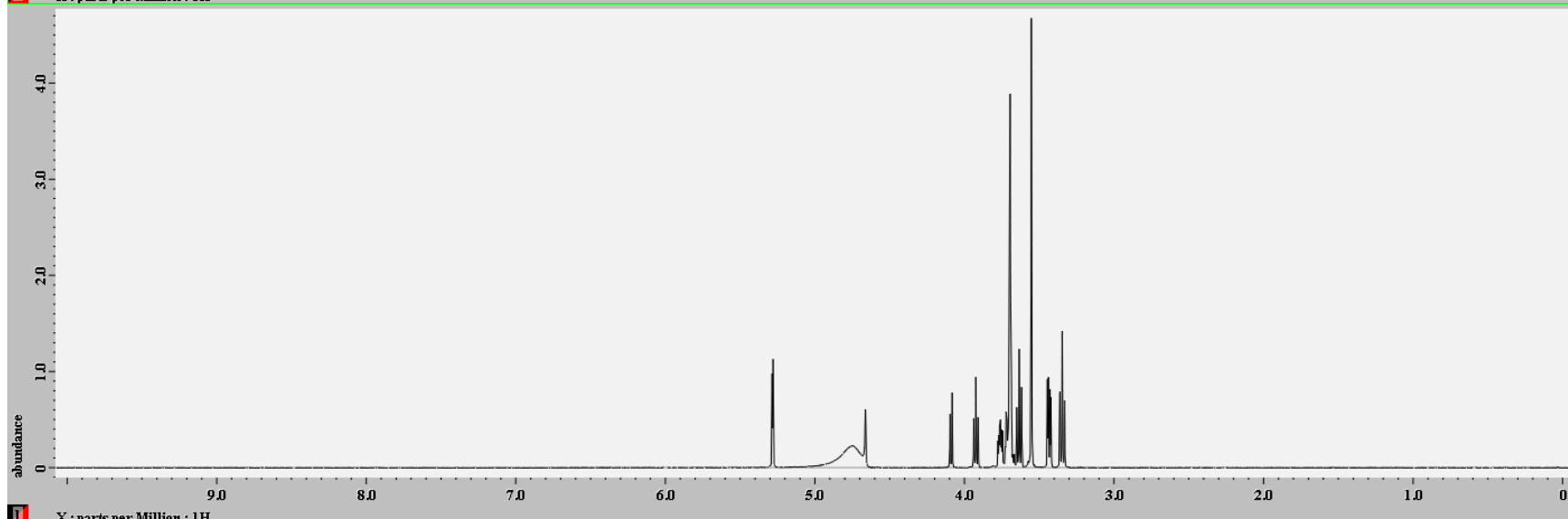
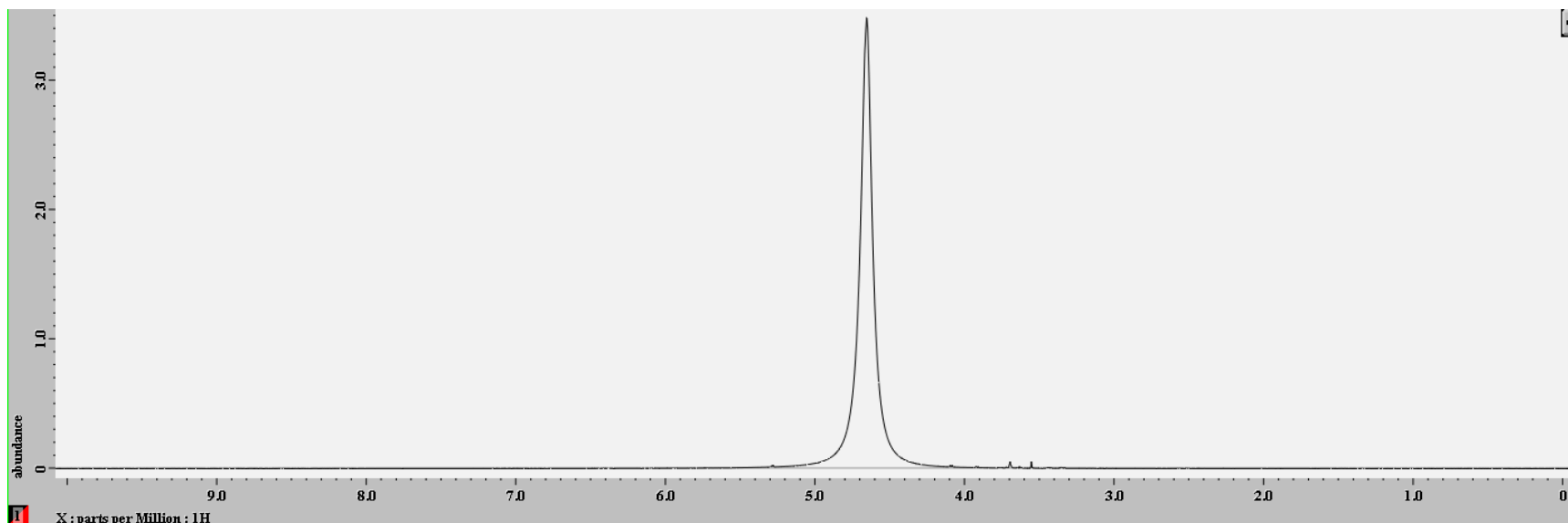
- Ideally NMR samples are dissolved in fully deuterated solvent, this provides the lock signal and also will not appear in the spectrum
- Large protonated signals will prevent the observation of small signals due to the dynamic range limitation of the spectrometer
- They may also obscure signals, contributing to  $T_1$ -noise and baseline distortions
- The solution to these problems is to remove the solvent signal

# Suppression methods

---

- There are many different ways available to suppress solvents, but fundamentally these fall into three categories:
  - One can presaturate the solvent resonance with weak, continuous rf at its frequency
  - Manipulate the solvent magnetization to produce zero magnitude (“jump and return”)
  - Use pulsed field gradients (WATERGATE, WET) to eliminate solvent resonances
- All methods work best when the sample is well shimmed!

# Pre-saturation to suppress a solvent signal



# Pre-saturation to suppress a solvent signal

Delta version 4 series

[ruji] Experiment Tool: single\_pulse.exe2

File Tools View Options

Header Instrument Acquisition Pulse

x\_angle 45[deg]

x\_90\_width 7.66[us] x90

x\_atn 3[dB]

x\_pulse 3.83[us]

relaxation\_delay 5[s]

repetition\_time 6.81992[s]

dante\_presat

presat\_time 5[s] relaxation\_delay

dante\_pulse 2[us]

dante\_interval 0.1[ms]

dante\_attenuator 40[dB]

dante\_loop 490

irr\_mode Presaturation

irr\_domain Proton

irr\_offset 4.63947 [ppm]

irr\_attenuator 40[dB]

tri\_mode Off

tri\_domain Proton

tri\_offset 5 [ppm]

tri\_attenuator 40[dB]

eca600 demo.pulask.com Total Collection Time: 00:01:00

Delta version 5 series

Spectrometer Control

Connection Tools Config Experiment

gradamp eca500

User: delta Owner: delta

Sample: Sample (1)  
Job: -  
Method: -  
Action: Idle  
Collected: -  
Time: -

Current tuning information for Probe is missing or incomplete.

Sample Name	Solvent	Slot	Kind	Preparation	Comment
Sample	Chloroform-D	1	Liquids	TRUE	

Open Jobs

- New Job 8
  - Proton
    - Proton 0:02

Header Instrument Acquisition Pulse Diagram Favorites

phs\_shft 0

irr\_mode Presaturation

irr\_domain Proton

irr\_offset 5 [ppm]

irr\_attenuator 40 [dB]

tri\_mode Off

tri\_domain Proton

tri\_offset 5 [ppm]

Deliver data automatically

Submit Job

Receiver Gain: 50 Spin: 0 [Hz] Lock: 884 Temp: 25 [dC] Helium: 50 [%] Nitrogen: 75 [%] Queue Length: 0

# Presaturation Methods

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- The basic presaturation is a long, low power, irradiation at the solvent position
- Better results can be obtained with methods that only excite the regions with good r.f. homogeneity, which excludes peripheral regions contributing to the water 'hump'
- Presaturation methods are
  - simple and reliable
  - suppress exchangeable protons

# Gradient solvent suppression methods

- These work by using pulsed field gradients acting to dephase the solvent signal.

WATERGATE (Water suppression by Gradient Tailored Excitation)

Good for protein and bio-molecule sample with water.  
Exchangeable protons are observable.

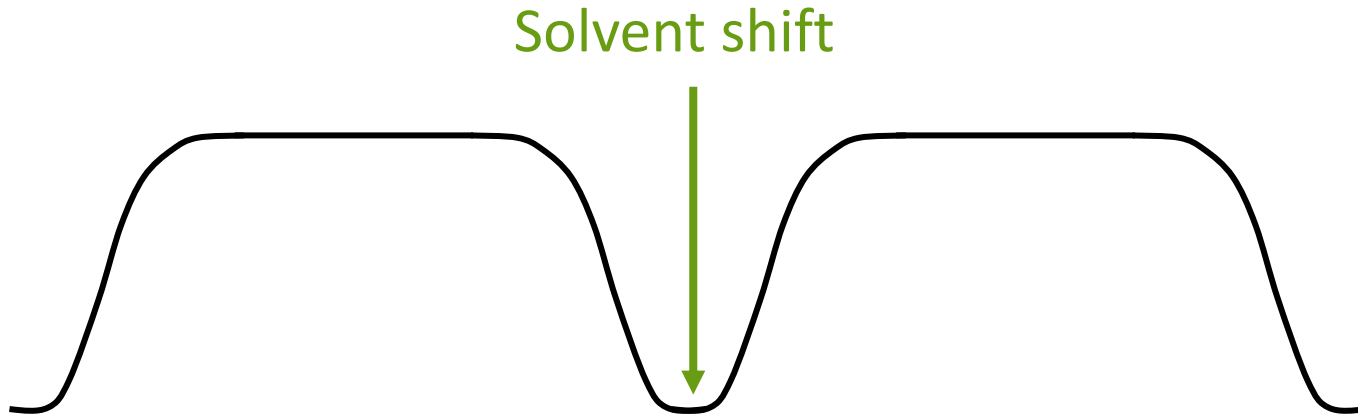
WET (Water suppression enhanced through  $T_1$  effect)

Because we can remove  $^{13}\text{C}$  satellite signals, this is good for organic solvent suppression (No-D NMR and LC-NMR).

# Watergate method

## Water Suppression by Gradient Tailored Excitation

Excitation profile



(Nonsolvent resonances near the solvent are also attenuated)

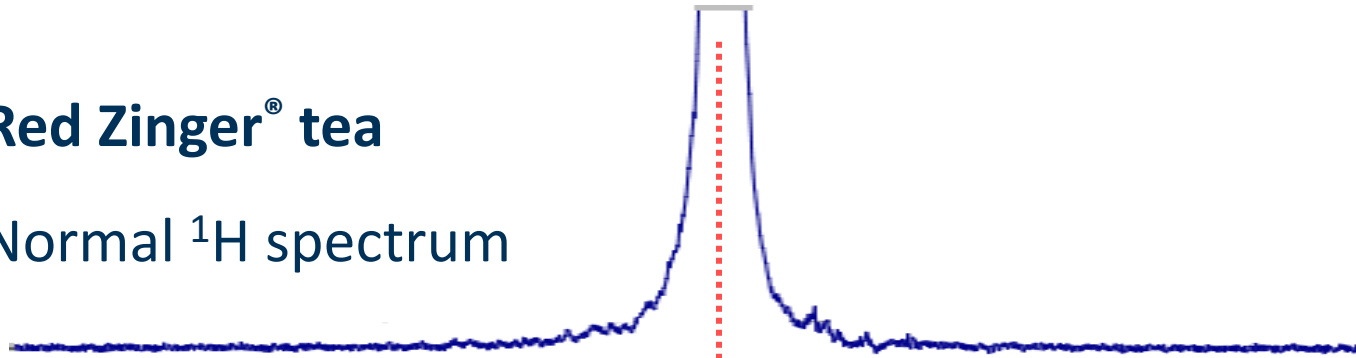
Experiment name in Delta: wgh.jxp (Delta V5), wgh.ex2 (Delta V4)



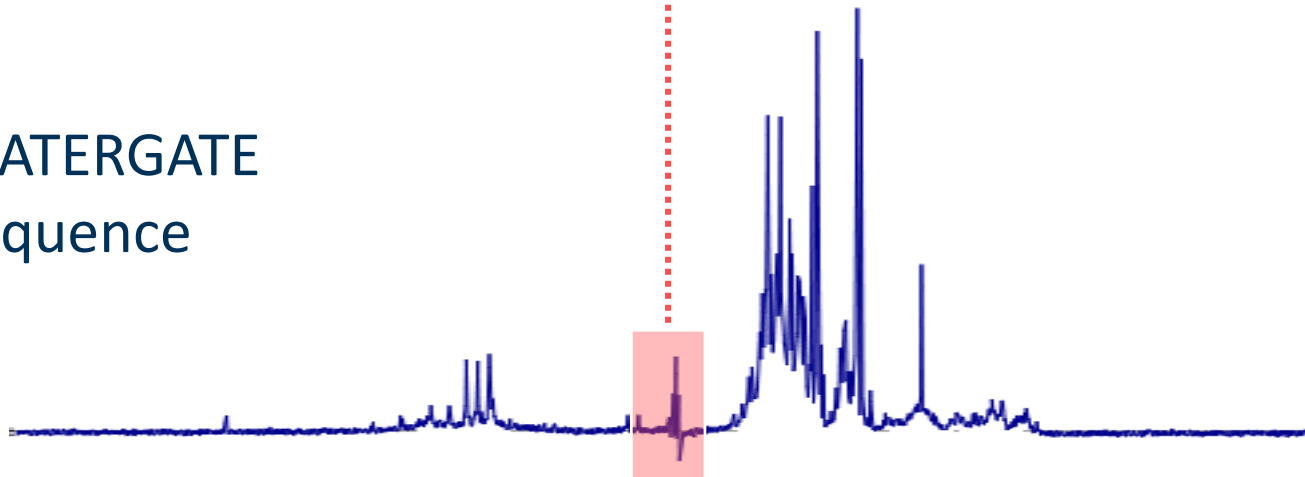
# Watergate of herbal tea

Red Zinger<sup>®</sup> tea

Normal <sup>1</sup>H spectrum

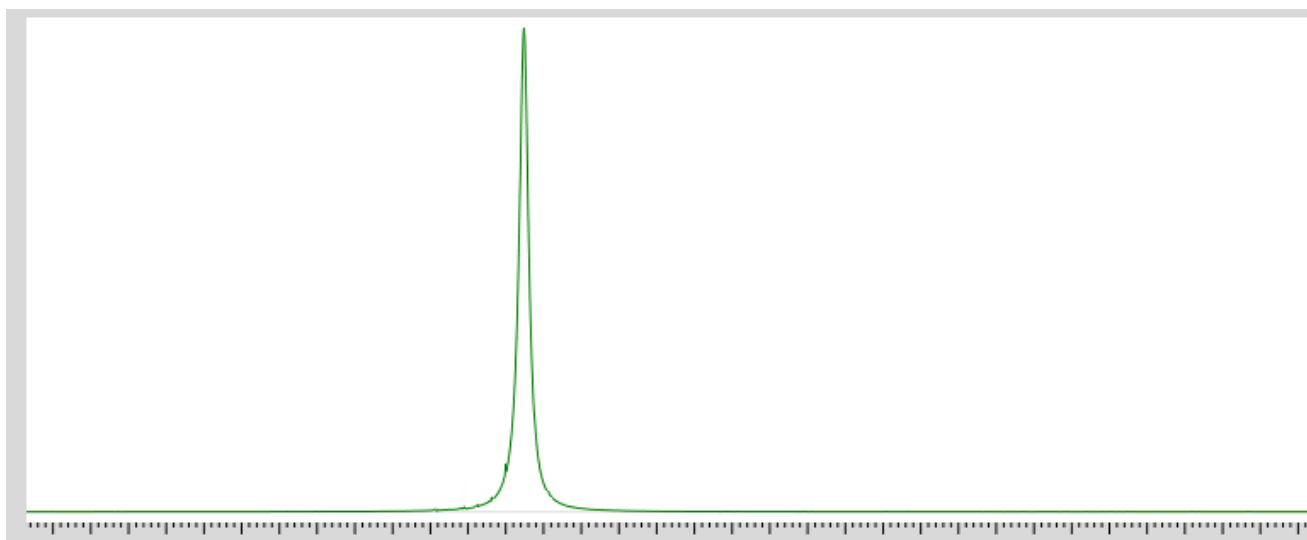


WATERGATE  
sequence



Residual H<sub>2</sub>O

# Robust5 method for solvent suppression

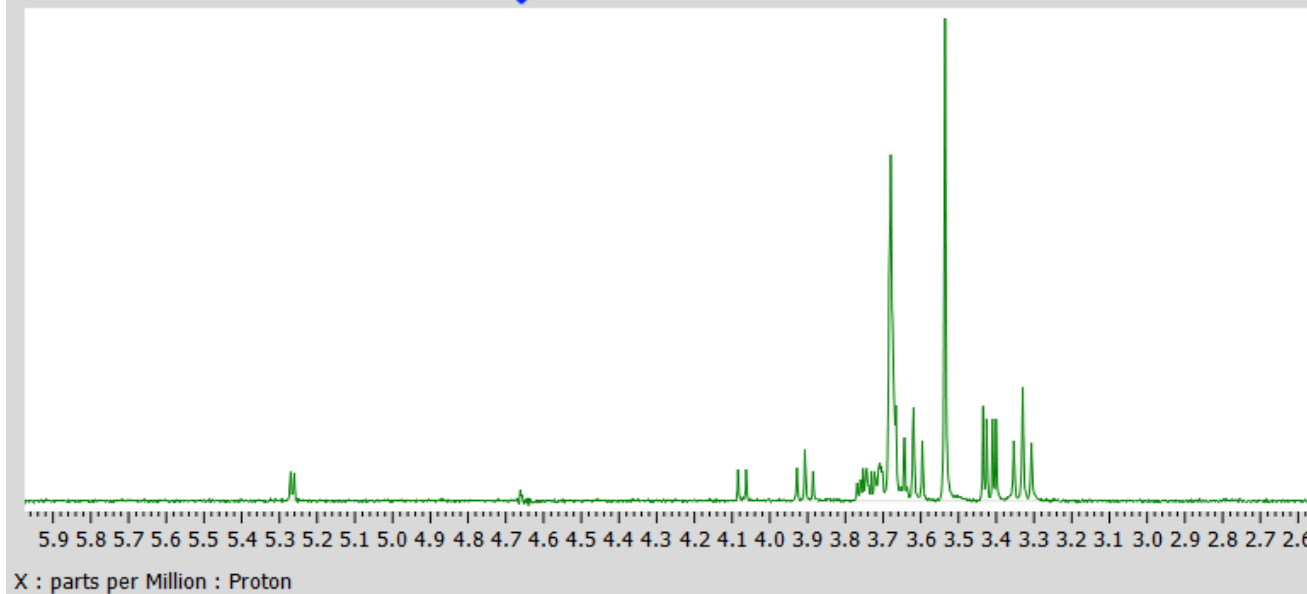


2mM sucrose  
sample  
90% H<sub>2</sub>O /  
10% D<sub>2</sub>O

Experiment  
based on perfect  
spin echo with  
180 pulses  
replaced by  
Watergate  
blocks

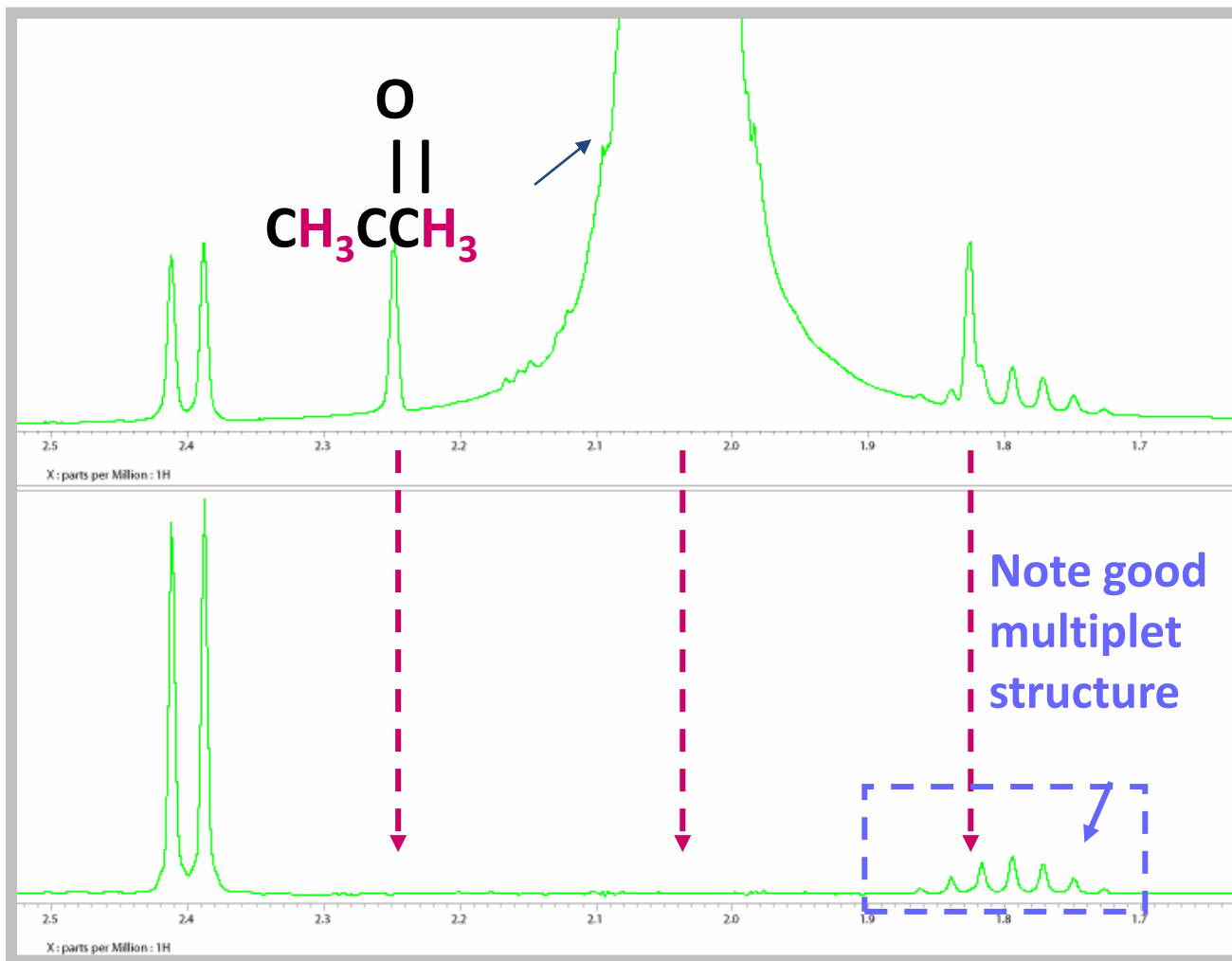


After 1 minute without any user intervention



X : parts per Million : Proton

# Wet result (use noD script)



Experiment name in Delta: single\_pulse\_wet\_slp.jxp (Delta V5),  
single\_pulse\_wet\_slp.ex2 (Delta V4)

**VT**

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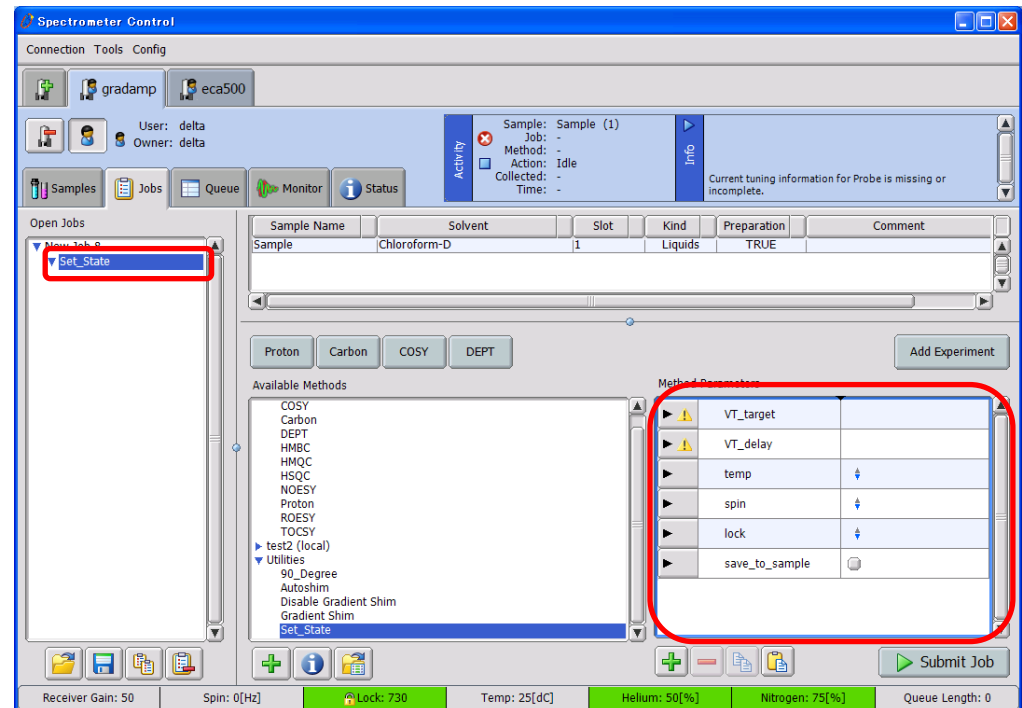
# VT in automation

- Delta can be configured to set a temperature for each sample:

Delta version 4 series



Delta version 5 series



# Low Temperature

- Delta can maintain a low temperature with FTS system at zero degrees C.
- Remember to change the switch in VT controller panel



# VT Nitrogen Dewar

- The VT dewar 'trunk' attaches to the probe heater. The probe heater acts in conjunction with the dewar to provide precise and stable low-temperature control by N<sub>2</sub> boil-off.



**VT Dewar Setup**

# VT Nitrogen dewar usage

- Check Handling of hardware manual

When the setting temperature is low, the nitrogen gas flowing in the probe may cause the sample to float. In such a case, attach the provided weight to the rotor. Use about three weights when the temperature is set to  $-75.1\text{ }^{\circ}\text{C}$  or less.

If there is trouble ejecting sample there are two eject air adjustments at the back of the console.

Eject valve is for when VT is used.

Eject-S is for when no VT is used.





# VT runs

- To run the same sample at a number of different temperatures
  - Note solvent freezing / boiling point
    - eg. CDCl<sub>3</sub> (m.p. -63.5C, b.p. +61.2C)
  - Only take VT to within 10-15C of these limits
    - eg. CDCl<sub>3</sub> min ca. -50C / max ca. +50C
  - Re-tune sample at ca. 30 degrees C intervals
  - Gradually change temperature to prevent thermal shock of the glass tube or the insert in the probe. I recommend no more than 20C change every 5 min. Array of VT\_proton methods allow this

# Thank you

You can discover more at

- <http://www.jeol.co.jp/en/>  
(Products -> NMR)
  - Description of our products
  - Free processing software
  - Free natural products database
  - Application notes
  - Events
  - And more

- <http://nmrsupport.jeol.com/> (license)

The screenshot displays the JEOL website interface. At the top, there is a navigation bar with the JEOL logo and links for PRODUCTS, APPLICATIONS NOTES, SUPPORT, and ABOUT US. Below this is a NEWS section with several articles dated from 2018/08/14 to 2018/08/24. A callout box titled 'Important information for NMR users' is also visible. The CASE STUDY section features three featured stories with images and brief descriptions of users from the University of Tokyo, DIC Corporation, and Toray Research Center. The PRODUCT LINEUP section shows images of various NMR spectrometers and software, including EC2R NMR spectrometer FT NMR, EC2S NMR spectrometer FT NMR, Delta NMR Software, Year Hold Magnet, and Magnet. At the bottom, there is a grid of eight informational boxes covering topics such as NMR data processing software, NMR peripherals/consumables, User stories, NMR basic knowledge/history, CH-NMR-NP, Liquid/solid state NMR probes, NMR application note, and quantitative NMR.