

NMR Training Course

9th September 2021 Adolfo Botana, PhD JEOL UK Demo Lab



Diffusion-Ordered SpectroscopY



Mixture analysis by NMR

applications

metabolomics drug development process chemistry food science natural products chemistry organic synthesis

pros/cons

- + structural information
- + nondestructive
- low(ish) sensitivity
- usually needs separation (e.g. LC-NMR)



Self-Diffusion



- Molecules experience both rotational and translational Brownian motion
- Stokes-Einstein equation (spherical molecules) can be used to determine the mobility of the molecules

 $D = \frac{k_{\rm B}T}{6\pi\eta r_{\rm H}}$

- $k_{\rm B}$: Boltzmann constant *T*: temperature η : viscosity
- $r_{\rm H}$: hydrodynamic radius
- The equation is valid for solute molecules at infinite dilution diffusing through a continuum solvent (i.e. where the solvent molecules are much smaller than the solute).

The diffusion coefficient

$$D = \frac{k_{\rm B}T}{6\pi\eta r_{\rm H}}$$

The hydrodynamic radius *a* is the effective average radius of the solvated solute molecules, and will depend on the molar mass *MW*. Assuming similar chemistries (i.e. constant density)

• for a spherical molecule such as a globular protein,

$$D\propto (MW)^{-1/3}$$

- for a 'random coil' polymer or a flat disk,
- for a rigid linear molecule

 $D \propto (MW)^{-1/2}$ $D \propto (MW)^{-1}$

In practice D will also depend on concentration, molecular shape, interactions etc.

Self-Diffusion and molecular weight

- Morris approximation for the correlation between molecular weight and selfdiffusion coefficient
- More advanced than previous estimation of $r_H \propto \sqrt[3]{MW}$
- Takes into account the molecular interactions of solvent and solutes
- Calculator available: Temperature, solvent and MW or D
- $k_{\rm B}$: Boltzmann constant
- T: temperature
- η : viscosity
- $ho_{\rm eff}$: effective density of the molecule (packing effects, geometry, solvation and flexibility)
- MW : molecular weight of the molecule
- $MW_{\rm S}$: molecular weight of the solvent
- N_A: Avogadro number



R. Evans et al, *Angew. Chem. Int. Ed.* (2013), 52: 3199–3202 R. Evans et al, *Anal. Chem.* (2018), *90* (6), 3987–3994 R. Evans, *Prog Nucl Magn Reson Spectrosc* (2020), 117, 33-69 <u>https://www.nmr.chemistry.manchester.ac.uk/?q=node/432</u>

Spin echo



https://twitter.com/wonderofscience/status/1320692387925032961

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Pulsed field gradient spin echo

Application of PFG will result in a diffusivity dependent ۲ attenuation 4.5 4 3.5 0.5 0 ppm 2.5 . 7 1.5 180° 90° $S = S_0 e^{-D\gamma^2 \delta^2 G^2 \Delta'}$ RF S: signal amplitude S_0 : signal amplitude without diffusion Λ D: diffusion coefficient γ : gyromagnetic ratio δ : gradient pulse width G G : gradient amplitude phases unscrambled spin phases scrambled Δ' : corrected diffusion time $[\boldsymbol{\Phi} = \boldsymbol{\Phi} (\boldsymbol{z}) - \boldsymbol{\Phi} (\boldsymbol{z}) = 0]$ $\left[\phi = \phi(z) \right]$ if spins haven't moved

Pulsed field gradient spin echo

• Magnetization evolution without diffusion



Pulsed field gradient spin echo

- Magnetization evolution with diffusion
- Diffusion results in imperfect refocusing of magnetization.
- More diffusion leads to worse refocusing, and thus more signal attenuation

No diffusion vs diffusion







Diffusion-Ordered SpectroscopY (DOSY)



- Measure spectra as a function of G
- Fit peak heights to get diffusion coefficients D

 $S = S_0 e^{-D\gamma^2 \delta^2 G^2 \varDelta'}$

- Extend 1D peaks into a second dimension, with Gaussian shapes centred on the *D*'s
- Widths determined by the standard errors σ_{D}

Virtual chromatography

Despite the name, DOSY is not like COSY/NOESY: spectra are statistical constructs from, not transforms of, experimental data

Practical consequences of the Stejskal-Tanner formula

Stejskal-Tanner formula for signal intensity:

$$S = S_0 e^{-D\gamma^2 \delta^2 G^2 \Delta'}$$

- ($\gamma \ \delta \ G_{zi}$)² - gradient area

nuclei with high γ values are more sensitive for diffusion (¹H, ¹⁹F, ³¹P) (i.e. ¹H is 16 times more sensitive than ¹³C)

- δ should be kept short

during δ the magnetization is transverse, homonuclear J-couplings evolve

- G the more, the better

provided the gradient hardware allows it

- Δ should be kept short:

to minimize convection effects

Why do we lose signal intensity?

1. Instrumental imperfections gradient system recovery - Eddy currents with multiple time constants phase and lineshape distortions gradient linearity and symmetry gradient reproducibility noise produced by the gradient amplifier Use suitable pulse sequence **2.** T₁ or T₂ relaxation Minimize pulse sequence delays 3. Diffusion - the physical basis of DOSY 4. Convection - the enemy of every diffusion experiment! Minimize sample volume Increase sample viscosity Minimize temperature gradients Use convection compensated sequences

PFG Spin echo

Field gradient pulses result in signal attenuation



J-modulated sexted



PFG Stimulated echo

- Magnetization is stored along the z-axis for most of Δ
- Reduced J-modulation
- Lose 50% of magnetization



In practice: Bpp_led_dosy_pfg.jxp



DOSY of 3 port wines



DOSY expansion of 3 years old port wine



J. Agric. Food Chem. **2004,** 52, 3736

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Convection

• Once a critical Raleigh number is reached (1700 for a Benard configuration, i.e., two parallel horizontal boundaries separated by a distance *d*), natural convection appears.



Convection in an NMR tube

Aromatic signals from quinine (7.1 to 7.6 ppm) as a function of increasing gradient strength at 25 $^\circ \rm C.$



Gradient amplitude

J. Magn. Reson. 177 Juio 203 (2005)

Convection in NMR

- Any pulse sequence with encoding-decoding gradient pairs suffers from convection
 - Gradient shimming (ask for alternative)
 - Gradient selective 1D
 - g....jxp
 - ...

Convection compensated diffusion measurement: bpp_dste_led_dosy_pfg.jxp

Double stimulated echo: lose 75% of signal and longer phase cycling



signal overlap in DOSY processing



signal overlap in DOSY processing



Superimposed exponentials is a very difficult mathematical problem (illposed and numerically unstable).

It is only practically feasible with high signal to noise ratio and for a limited (2-3) number of exponentials.





Overlapping peaks give compromise diffusion coefficient

Anal. Chem. 2006, 78, 3040-3045



 $D_{\rm A}$ and $D_{\rm B}$ must differ by at least 30% Very dependent on the quality and S/N of data

Anal. Chem. 2006, 78, 3040-3045

Effects of overlap with J-modulation



https://doi.org/10.1016/j.jmr.2010.11.012

signal overlap in DOSY processing



Here we fit the whole spectrum at once rather than each peak at the time as in HR-DOSY. This makes sense as all peaks in a component spectrum decays in the same way



$\mathbf{X} = \mathbf{C} \, \mathbf{S}^{\mathsf{T}} + \mathbf{E}$

Minimize E assuming a known decay form

SCORE: $\mathbf{E} = \mathbf{X} - \mathbf{CS}^{\mathsf{T}}$ (residuals) OUTSCORE: $\mathbf{E} = |\mathbf{S}_i| \cdot |\mathbf{S}_j|$ (spectral similarity)

Chem. Commun. 2013, 49, 10510; Anal. Chem. 2008, 80, 3777

a mixture of progesterone and estradiol in DMSO- d_6 10-10⁻¹⁰m²s-1 HRDOSY 2.5 1.5 0.5 SCORE D = 8.55 D = 12.2 2.5 1.5 0.5 OUTSCORE D = 9.03D = 10.89 2.5 1.5 d/ppm

HRDOSY

monoexponential fitting <1% difference in D suffers from overlap

SCORE

minimizes residuals >30% difference in D

OUTSCORE

minimizes cross-talk <5% difference in D fewer components

Chem. Commun. 2013, 49, 10510

Multivariate analysis



Figure 1. Schematic illustration of the FIDDLE algorithm for reference deconvolution. The reference peak is extracted from the experimental spectrum (b), and its inverse Fourier transform (c) is compared to that of "perfect" FID (e) to yield a correction function (e/c). The correction is then applied in the time domain to the entire experimental FID (a) to produce the corrected FID (f).

• Example (3 compounds with varying concentrations):



Multivariate DOSY NMR of paracetamol tablet



Solvent Methanol-d₄

signal overlap in DOSY processing



overlap in 2D DOSY: ¹³C DOSY

500 MHz ¹H and ¹³C DOSY spectra of mixture of alcohols in D₂O



No overlap in the ¹³C spectrum greatly facilitates interpretation

https://doi.org/10.1016/j.jmr.2011.03.016

600 MHz ¹H and ¹⁹F DOSY spectra of fluorinated compounds in DMSO-d₆



Magn. Reson. Chem. 2014 52, 172

DOSY can only separate signals from species that have different diffusion coefficients



Can we manipulate the way different species diffuse?

interaction of a solute with a [more slowly diffusing] matrix reduces its apparent diffusion in proportion to the strength of interaction.

a simple two-site model $D_{apparent} = f_{bound} D_{matrix} + (1 - f_{bound}) D_{free}$ fast exchange bound molecules molecules in free solution

isomers resolved using micelles



using a micellar matrix exploits differences in binding to separate the signals of species with similar or identical diffusion coefficients Anal. Chem. 2009, 81, 4548

Chiral MAD: epimers resolved using cyclodextrins



differential inclusion by β -cyclodextrin of the epimers of the natural product naringin is exploited, allowing separation of the naringin epimer signals by high resolution DOSY.

Lanthanide shift reagents



an "impossible" mixture of hexane, hexanol and hexanal.

adding Eu(fod)₃ resolves the signals in both dimensions. The signals from hexane, hexanal and hexanol can now be identified

Chem. Commun. **2011**, 47, 7063

Thank you

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 - Description of our products
 - Free processing software
 - Free natural products database
 - Application notes
 - Events
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